

On the use of natural orbitals in GRASP calculations

Per Jönsson

Department of Materials Science and Applied Mathematics
Malmö University, Sweden

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Dedicated to the memory Ian and Charlotte



Acknowledgment

- ▶ Betul Atalay
- ▶ Gediminas Gaigalas
- ▶ Michel Godefroid
- ▶ Jörgen Ekman
- ▶ Jacek Bieron
- ▶ Jiguang Li
- ▶ Sacha Schiffmann

Starting point

- ▶ I think we are orbital basis limited
- ▶ crucial that the orbitals in the MR are as accurate as possible
- ▶ the CSFs in the MR should be described by orbitals that are mutually non-orthogonal: example Ti I: $3d^24s^2\ ^3F_{2,3,4}$ and $3d^3(^4F)4s\ ^5F_{1,2,3,4,5}$.
- ▶ multiple and non-orthogonal orbitals targeted for different pair correlation effects
- ▶ the number of allowed orbitals in GRASP should be increased
- ▶ transformation to natural orbitals

Independently Optimized Orbital Sets in GRASP—The Case of Hyperfine Structure in Li I

by Yanting Li^{1,2,†} , Per Jönsson^{2,*} , Michel Godefroid³ , Gediminas Gaigalas⁴ ,
Jacek Bieroń⁵ , José Pires Marques⁶ , Paul Indelicato⁷  and Chongyang Chen¹

¹ Shanghai EBIT Lab, Key Laboratory of Nuclear Physics and Ion-Beam Application, Institute of Modern Physics, Department of Nuclear Science and Technology, Fudan University, Shanghai 200433, China

² Department of Materials Science and Applied Mathematics, Malmö University, SE-20506 Malmö, Sweden

³ Spectroscopy, Quantum Chemistry and Atmospheric Remote Sensing, Université Libre de Bruxelles, B-1050 Brussels, Belgium

⁴ Institute of Theoretical Physics and Astronomy, Vilnius University, LT-010222 Vilnius, Lithuania

⁵ Instytut Fizyki Teoretycznej, Uniwersytet Jagielloński, 30-348 Kraków, Poland

⁶ LIP-Laboratório de Instrumentação e Física Experimental de Partículas and Faculdade de Ciências, Universidade de Lisboa, 1749-016 Lisboa, Portugal

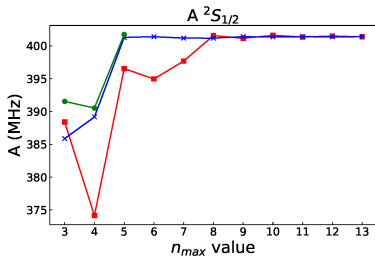
⁷ Laboratoire Kastler Brossel, Sorbonne Université, CNRS, ENS-PSL Research University, Collège de France, Case 74, 4, Place Jussieu, 75005 Paris, France

* Author to whom correspondence should be addressed.

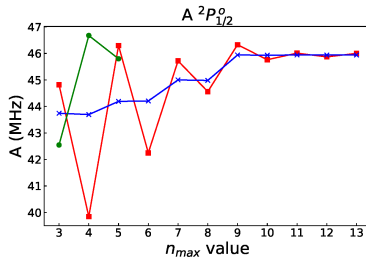
† These authors contributed equally to this work.

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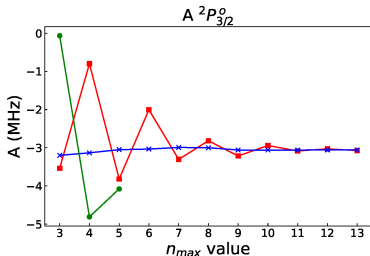
Intimidation – hyperfine structure



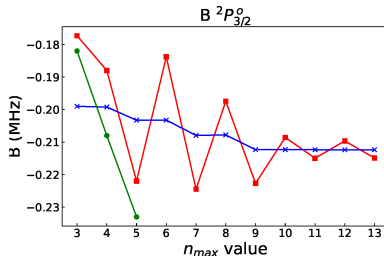
(a)



(b)



(c)



(d)



Coulomb (Velocity) Gauge Recommended in Multiconfiguration Calculations of Transition Data Involving Rydberg Series

by Asimina Papoulia^{1,2,*} , Jörgen Ekman¹ , Gediminas Gaigalas³ , Michel Godefroid⁴ , Stefan Gustafsson¹ , Henrik Hartman¹ , Wenxian Li¹ , Laima Radžiūtė³ , Pavel Rynkun³ , Sacha Schiffmann^{2,4} , Kai Wang⁵  and Per Jönsson¹ 

¹ Department of Materials Science and Applied Mathematics, Malmö University, SE-20506 Malmö, Sweden

² Division of Mathematical Physics, Department of Physics, Lund University, SE-22100 Lund, Sweden

³ Institute of Theoretical Physics and Astronomy, Vilnius University, Saulėtekio av. 3, LT-10222 Vilnius, Lithuania

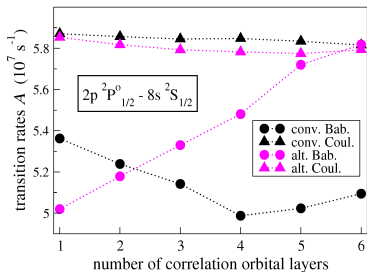
⁴ Chimie Quantique et Photophysique, Université libre de Bruxelles, B-1050 Brussels, Belgium

⁵ Hebei Key Lab of Optic-electronic Information Materials, College of Physics Science and Technology, Hebei University, Baoding 071002, China

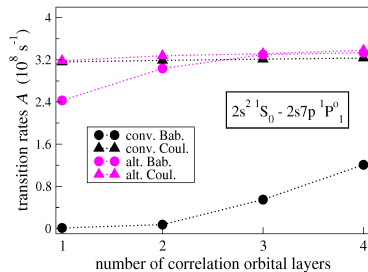
* Author to whom correspondence should be addressed.

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Intimidation – transition rates



(a)



(b)

Outline

- ▶ transformation to natural orbitals in two electron systems
- ▶ computation of natural orbitals in the general case
- ▶ Sacha Schiffmann's analysis for hfs
- ▶ applications

$1s^2\ ^1S$ in He I – the beginning

- ▶ $\Psi^{l=0}(1s^2\ ^1S) = \sum_{n,n'} c_{nn'} \Phi(nsn's)$
- ▶ $\Psi^{l=0}(1s^2\ ^1S) = \left(\sum_{n,n'} c_{nn'} R_{ns}(r_1) R_{n's}(r_2) \right) |ss'\ ^1S\rangle$,
where $R_{nl}(r) = P_{nl}(r)/r$ and $c_{nn'} = c_{n'n}$.
- ▶ $\left(\sum_{n,n'} c_{nn'} R_{ns}(r_1) R_{n's}(r_2) \right) = \mathbf{R}(\mathbf{r}_1) \mathbf{C} \mathbf{R}(\mathbf{r}_2)^t$,
radial factor in matrix vector form
- ▶ $\mathbf{C}$ symmetric, unitary transformation that will diagonalize it
- ▶ applying transformation to radial function vectors $\mathbf{R}$ gives the natural orbitals (NO)
- ▶ $\Psi^{l=0}(1s^2\ ^1S) = \sum_n c_n \Phi(ns^2)$
- ▶ similar transformations for other l , leads to reduced form
 $\Psi(1s^2\ ^1S) = \sum_l \sum_n c_{nl} \Phi(nl^2)$



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Relativistic radial electron density functions and natural orbitals from GRASP2018 ☆, ☆☆

[S. Schiffmann](#)^{a, b} , [J.G. Li](#)^c, [J. Ekman](#)^d, [G. Gaigalas](#)^e, [M. Godefroid](#)^a , [P. Jönsson](#)^d, [J. Bieroń](#)^f

- NOs are obtained, for each κ , by constructing and diagonalizing the orbital electron density matrix

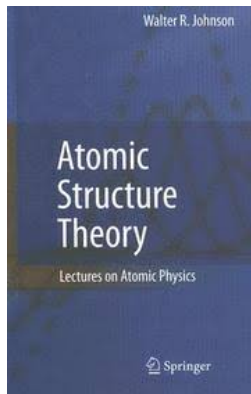
Effects of natural orbitals

- ▶ when CV correlation are included in fully variational calculations, valence orbitals contracts relative to their shapes from DF
- ▶ in the layer-by-layer approach the valence orbitals are fixed from the DF calculation, no contraction
- ▶ transformation to NO from layer-by-layer wave functions leads to a contraction of the valence orbitals similar to the one from fully variational calculations
- ▶ important implications for hfs, IS and transition rates

Effects of natural orbitals

- ▶ building CSF expansions on NOs concentrates the weights to fewer CSFs
- ▶ more compact representation, we can condense harder
- ▶ energies lower in NO basis compared to the ordinary basis
- ▶ some terms in Rayleigh-Schrödinger perturbation theory are zero in the NO basis

Natural orbitals vs Brueckner orbitals



- ▶ Similarities with Brueckner orbitals that have a clearer physical interpretation, read section 7.3.3

Transformation to natural orbitals - Na I

PHYSICAL REVIEW A

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Natural orbitals in multiconfiguration calculations of hyperfine-structure parameters

Sacha Schiffmann, Michel Godefroid, Jörgen Ekman, Per Jönsson, and Charlotte Froese Fischer
Phys. Rev. A **101**, 062510 – Published 5 June 2020

Computational procedure

- ▶ $2s^22p^63s\ ^2S_{1/2}$ and $2s^22p^63p\ ^2P_{1/2,3/2}$ in Na I.
- ▶ layer-by-layer optimization of correlation orbitals based on SD-CV and SD-CV-CC expansions
- ▶ transforming to NOs based on SD-CV expansion
- ▶ CI based on SD-CV-CC + TQ expansion in NO basis
- ▶ detailed comparisons with corresponding fully variational as well as layer-by-layer MCHF calculations

► Orbital contraction

$$\langle r_{3s} \rangle_{\text{LBL}} = 4.12007 \rightarrow \langle r_{3s} \rangle_{\text{NO}} = 4.05088$$

$$\langle r_{3s}^{-2} \rangle_{\text{LBL}} = 0.43488 \rightarrow \langle r_{3s}^{-2} \rangle_{\text{NO}} = 0.48120$$

Hyperfine structure

Active set	$A_{1/2}(\text{MHz})$	
	LBL	NO
DHF	633.698	
MCDHF+CI CV		
3	691.693	
4	837.150	
5	870.354	
6 <i>h</i>	895.195	
7 <i>h</i>	906.639	
8 <i>h</i>	939.435	
9 <i>h</i>	938.813	937.083
MCDHF+CI CV + CC		
10 <i>h</i>	852.679	888.676
11 <i>h</i>	852.806	888.725
CI CV + CC + T		
SD[11 <i>h</i>]		
∪ T[4]	859.307	886.605
∪ T[5 <i>f</i>]	865.388	885.925
∪ T[6 <i>f</i>]	866.826	883.113
CI CV + CC + T + Q		
SD[11 <i>h</i>]		
∪ TQ[4]	862.146	889.111
∪ T[6 <i>f</i>] ∪ Q[4]	869.945	885.841
Expt.	885.813 064 4(5) [34]	

Hyperfine structure

TABLE VIII. Relativistic magnetic dipole hyperfine constants $A_{1/2}$ (MHz) and $A_{3/2}$ (MHz) of the $^2P_{1/2}$ and $^2P_{3/2}$ sodium excited states, respectively, along the active space expansion in two different orbital bases. Similar results are given for the electric field gradient $EFG = B_{3/2}/Q$ (MHz/b) of the $J = 3/2$ level.

Active set	$A_{1/2}$ (MHz)		$A_{3/2}$ (MHz)		$B_{3/2}/Q$ (MHz/b)	
	LBL	NO	LBL	NO	LBL	NO
DHF	64.157		12.744		15.939	
MCDHF+CI CV						
3	69.348		12.054		16.605	
4	91.212		20.611		27.648	
5f	93.643		21.077		27.291	
6f	98.294		20.300		27.822	
7f	100.751		20.770		28.520	
8f	101.791		20.544		28.562	
9f	101.861	101.817	20.554	20.543	28.453	28.433
MCDHF+CI CV + CC						
10f	87.740	94.655	17.184	18.592	24.424	26.274
11f	87.701	94.603	17.197	18.606	24.469	26.322
CI CV + CC + T						
SD[11f]						
∪ T[4]	88.596	94.668	17.514	18.755	24.838	26.474
∪ T[5f]	90.073	94.594	17.943	18.874	25.395	26.616
∪ T[6f]	90.908	94.316	18.083	18.803	25.675	26.599
Others						
CI [1]	94.04		18.80		25.79	
SD [38]	94.99		18.84		26.85	
SD [37]	92.4		19.3			
CCSD [36]	93.02		18.318		26.14	
Expt. [43,44]	94.42(19)		18.79(12)			

- ▶ Computing along the isoelectronic sequence shows that the effect of the NOs are largest for neutral and near neutral systems

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**Astronomy
&
Astrophysics**

MCDHF and RCI calculations of energy levels, lifetimes, and transition rates in Si III and Si IV[★]

B. Atalay^{1,2}, T. Brage^{1,4}, P. Jönsson³, and H. Hartman³

¹ Department of Physics, Lund University, Post Office Box 118, 22100 Lund, Sweden
e-mail: betul.atalay@teorfys.lu.se

² Department of Physics, Çanakkale Onsekiz Mart University, Çanakkale, Turkey
e-mail: batalay@comu.edu.tr

³ Materials Science and Applied Mathematics, Malmö University, 20506 Malmö, Sweden

⁴ Institute of Modern Physics, Fudan University, Shanghai, PR China

Computational procedure

- ▶ $2s^2 2p^6 nl$, $n \leq 7$ and $l \leq 6$, configurations in Si IV.
- ▶ we performed the calculations for states belonging to $2s^2 2p^6 nl$, $n \leq 9$ and $l \leq 6$, i.e., we added two more layers spectroscopic orbitals in comparison to the states we were targeting.
- ▶ two layers aim to correcting the tail of the wave functions for the targeted states
- ▶ optimize correlation orbitals on MR-SD-CV
- ▶ optimize additional correlation orbitals on the $1s^2 2s^2 2p^6$ core based on SD-CC expansion
- ▶ transform to NOs based on SD-CC expansion
- ▶ CI based on MR-SD-CV-CC expansion in NO basis

Compactness of wave function

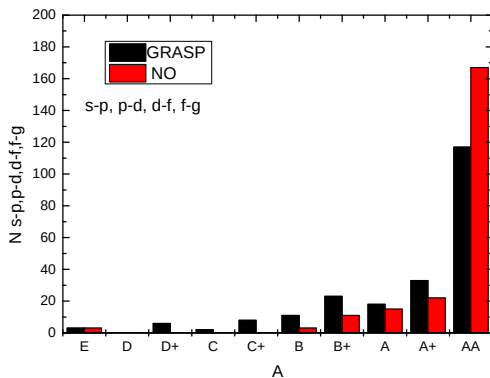
- ▶ energy level structure unchanged
- ▶ expansion concentrated to fewer CSFs
- ▶ take the expansion and accumulate the contributions up to 0.999999 of the full wave function. See how many CSFs survive

	ordinary	NO
core	6163/19253	3710/19253
even parity	437 703/995 020	264 047/995 020
odd parity	425 451/993 501	255 974/993 501

Transition rates

- ▶ results from original calculation very good
- ▶ relative difference between transition rates in length and velocity gauge is **halved** in NO basis
- ▶ rates in velocity gauge more stable with respect to the NO basis, i.e. it is the length gauge that is mostly affected
- ▶ Gediminas QQE analysis shows that transitions are promoted to more accurate classes according to NIST definition
- ▶ NIST classes: $AA \leq 1 \%$, $A^+ \leq 2 \%$, $A \leq 3 \%$, $B^+ \leq 7 \%$, $B \leq 10 \%$, $C^+ \leq 18 \%$, $C \leq 25 \%$, $D^+ \leq 40 \%$, $D \leq 50 \%$, and $E > 50 \%$

Transition rates



Transformation to natural orbitals - In I

Results due to Betul Atalay who unfortunately could not participate

- ▶ states of the $5s^25p$, $5s^26p$, $5s^27p$, $5s^28p$, $5s^24f$ odd and $5s^26s$, $5s^27s$, $5s^28s$, $5s^25d$, $5s^26d$, $5s5p^2$ even configurations.
- ▶ Hybrid strategy: large MR including also $5p^3$, $5p^26p$ etc

layer	orbitals	
1	$\{10s, 10p, 7d, 5f, 5g\}$	CV $5s^2$ as the core
2	$\{11s, 11p, 8d, 6f, 6g\}$	CV $5s^2$ as the core
3	$\{12s, 12p, 9d, 7f, 7g, 6h\}$	VV $4s^24p^64d^{10}$ as core
4	$\{13s, 13p, 10d, 8f, 8g, 7h, 7i\}$	VV $4s^24p^64d^{10}$ as core
5	$\{14s, 14p, 11d, 9f, 9g, 8h, 8i\}$	VV $4s^24p^64d^{10}$ as core
6	$\{15s, 15p, 12d, 10f, 10g, 9h, 9i\}$	VV and CV $4s^24p^64d^{10}$ as core
7	$\{16s, 16p, 13d, 11f, 11g, 10h, 10i\}$	VV and CV $4s^24p^64d^{10}$ as core
8	$\{17s, 17p, 14d, 12f, 12g, 10h, 10i\}$	VV and CV $4s^24p^64d^{10}$ as core

Computational procedure

CI calculation

- ▶ SDMR-VVCV in hybrid orbital basis plus SMR with $2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10}$ as core
- ▶ SDMR-VVCV in natural orbital basis plus SMR with $2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10}$ as core
- ▶ transformation to NOs based on the SMR expansion

Transformation to NOs based on different types of expansions in the three examples. Why?

- ▶ "Research is what I'm doing when I don't know what I'm doing"
- ▶ inclination is that the SMR expansion is the one to use for transformation to NOs

Energies in cm^{-1}

State	layer8+S	layer 8+S+NO	NIST
$5s^2 5p^2 P_{1/2}^o$	0	0	0
$5s^2 5p^2 P_{3/2}^o$	2 222	2 230	2 212.599
$5s^2 6s^2 S_{1/2}$	24 354	24 265	24 372.957
$5s^2 6p^2 P_{1/2}^o$	31 785	31 703	31 816.982
$5s^2 6p^2 P_{3/2}^o$	32 079	31 997	32 115.251
$5s^2 5d^2 D_{3/2}$	32 995	32 827	32 892.230
$5s^2 5d^2 D_{5/2}$	33 013	32 846	32 915.539
$5s 5p^2 \ ^4P_{1/2}$	35 441	35 228	34 977.678
$5s 5p^2 \ ^4P_{3/2}$	36 498	36 277	36 020.780
$5s^2 7s^2 S_{1/2}$	36 239	36 147	36 301.864
$5s 5p^2 \ ^4P_{5/2}$	37 962	37 727	37 451.962
$5s^2 7p^2 P_{1/2}^o$	38 797	38 712	38 861.43
$5s^2 7p^2 P_{3/2}^o$	38 907	38 822	38 972.90
$5s^2 6d^2 D_{3/2}$	39 124	38 931	39 048.53
$5s^2 6d^2 D_{5/2}$	39 163	38 985	39 098.38
$5s^2 4f^2 F_{7/2}^o$	39 632	39 558	39 707.59
$5s^2 4f^2 F_{5/2}^o$	39 632	39 558	39 707.59
$5s^2 8s^2 S_{1/2}$	40 563	40 478	40 636.98
$5s^2 8p^2 P_{1/2}^o$	41 752	41 670	41 827.10
$5s^2 8p^2 P_{3/2}^o$	41 805	41 723	41 881.44
$5s^2 9s^2 S_{1/2}$	42 640	42 561	42 719.02
$5s^2 9p^2 P_{1/2}^o$	43 289	43 211	43 369.09
$5s^2 9p^2 P_{3/2}^o$	43 319	43 241	43 399.53
NSCFs	21 555 509	21 555 509	

Things to note

- ▶ $5s^25d$ and $5s5p^2$ states that were too high are brought down.
- ▶ ground state energy lowered from
–5873.7598418 H to –5874.0479647 H.
gain HUGE compared to what we gain by adding layers of
correlation orbitals!

Important paper

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Laser spectroscopy of indium Rydberg atom bunches by electric field ionization

[A. R. Vernon](#) , [C. M. Ricketts](#), [J. Billowes](#), [T. E. Cocolios](#), [B. S. Cooper](#), [K. T. Flanagan](#), [R. F. Garcia Ruiz](#), [F. P. Gustafsson](#), [G. Neyens](#), [H. A. Perrett](#), [B. K. Sahoo](#), [Q. Wang](#), [F. J. Waso](#) & [X. F. Yang](#)

[Scientific Reports](#) **10**, Article number: 12306 (2020) | [Cite this article](#)

Hyperfine structure, A constants

State	8 + S	8 + S NO	Saf	Sahoo	Sahoo exp	Exp
$5s^2 5p^2 P_{1/2}^o$	2417	2455	2306	2274	2282.04	2282
$5s^2 5p^2 P_{3/2}^o$	248.2	258.5	262.4	253	241.98	
$5s^2 6s^2 S_{1/2}$	1640	1702	1812	1645	1684.75	1685
$5s^2 6p^2 P_{1/2}^o$	249.7	280.7	263.2			250.2
$5s^2 6p^2 P_{3/2}^o$	62.96	71.94	77.82			79.33
$5s^2 5d^2 D_{3/2}$	-54.73	-62.85	-11.48	-9.74	-64	62
$5s^2 5d^2 D_{5/2}$	125.3	150.6	47.83	39.87	151.2	152
$5s 5p^2^4 P_{1/2}$	9512	9471				9194
$5s 5p^2^4 P_{3/2}$	4146	4132				3999
$5s^2 7s^2 S_{1/2}$	486.7	522.7	544.5	520		
$5s 5p^2^4 P_{5/2}$	3814	3791			3696	3654
$5s^2 7p^2 P_{1/2}^o$	85.37	100.2	95.61			90.7
$5s^2 7p^2 P_{3/2}^o$	23.02	27.40	30.83			32.3
$5s^2 6d^2 D_{3/2}$	-41.68	-63.59	-11.2			
$5s^2 6d^2 D_{5/2}$	137.45	204.4	30.81			
$5s^2 4f^2 F_{7/2}^o$	0.087	0.084	0.2293			
$5s^2 4f^2 F_{5/2}^o$	0.1243	0.142	0.1871			
$5s^2 8s^2 S_{1/2}$	199.3	220.1	240.8	233	243.85	
$5s^2 8p^2 P_{1/2}^o$	37.78	45.70	45.97			44.2
$5s^2 8p^2 P_{3/2}^o$	10.46	12.85	15.42			16.3

Transition rates - NIST classes

NIST classes: $AA \leq 1 \%$, $A^+ \leq 2 \%$, $A \leq 3 \%$, $B^+ \leq 7 \%$, $B \leq 10 \%$, $C^+ \leq 18 \%$, $C \leq 25 \%$, $D^+ \leq 40 \%$, $D \leq 50 \%$, and $E > 50 \%$

Transition rates - ordinary orbitals

```

1-5873.75984180  2s(2).2p(6).3s(2).3p(6).3d(10).4s(2).4p(6).4d(10).5s(2).5p_2P
1-5873.64887616  2s(2).2p(6).3s(2).3p(6).3d(10).4s(2).4p(6).4d(10).5s(2).6s_2S
24354.14 CM-1    4106.08 ANG(S(VAC))    4105.65 ANG(S(AIR))
E1  S =  3.42574D+00  GF =  2.53426D-01  AKI =  5.01313D+07  dT =  0.05054
      3.25260D+00      2.40618D-01      4.75977D+07  NIST class=B+

1-5873.64887616  2s(2).2p(6).3s(2).3p(6).3d(10).4s(2).4p(6).4d(10).5s(2).6s_2S
1-5873.61501516  2s(2).2p(6).3s(2).3p(6).3d(10).4s(2).4p(6).4d(10).5s(2).6p_2P
7431.63 CM-1     13456.00 ANG(S(VAC))    13454.60 ANG(S(AIR))
E1  S =  3.81661D+01  GF =  8.61562D-01  AKI =  1.58696D+07  dT =  0.03090
      3.69867D+01      8.34938D-01      1.53792D+07  NIST class=B+

1-5873.64887616  2s(2).2p(6).3s(2).3p(6).3d(10).4s(2).4p(6).4d(10).5s(2).6s_2S
1-5873.58306533  2s(2).2p(6).3s(2).3p(6).3d(10).4s(2).4p(6).4d(10).5s(2).7p_2P
14443.81 CM-1    6923.38 ANG(S(VAC))    6922.66 ANG(S(AIR))
E1  S =  5.36848D-01  GF =  2.35536D-02  AKI =  1.63883D+06  dT =  0.17660
      4.42041D-01      1.93941D-02      1.34941D+06  NIST class=C

1-5873.75984180  2s(2).2p(6).3s(2).3p(6).3d(10).4s(2).4p(6).4d(10).5s(2).5p_2P
1-5873.59472037  2s(2).2p(6).3s(2).3p(6).3d(10).4s(2).4p(6).4d(10).5s(2).7s_2S
36239.96 CM-1    2759.38 ANG(S(VAC))    2759.09 ANG(S(AIR))
E1  S =  2.68608D-01  GF =  2.95686D-02  AKI =  1.29514D+07  dT =  0.06453
      2.51275D-01      2.76606D-02      1.21157D+07  NIST class=B+

1-5873.61501516  2s(2).2p(6).3s(2).3p(6).3d(10).4s(2).4p(6).4d(10).5s(2).6p_2P
1-5873.59472037  2s(2).2p(6).3s(2).3p(6).3d(10).4s(2).4p(6).4d(10).5s(2).7s_2S
4454.19 CM-1     22450.76 ANG(S(VAC))    22448.44 ANG(S(AIR))
E1  S =  3.96306D+01  GF =  5.36196D-01  AKI =  3.54792D+06  dT =  0.06663
      3.69902D+01      5.00472D-01      3.31154D+06  NIST class=B

```

Transition rates - NOs

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3-5874.03780394 2s(2).2p(6).3s(2).3p(6).3d(10).4s(2).4p(6).4d(10).5s(2).5p_2P
1-5873.93740296 2s(2).2p(6).3s(2).3p(6).3d(10).4s(2).4p(6).4d(10).5s(2).6s_2S
22035.47 CM-1 4538.14 ANGSA(VAC) 4537.66 ANGSA(AIR)
E1 S = 8.28790D+00 GF = 5.54742D-01 AKI = 8.98355D+07 dT = 0.02105
      8.11343D+00      5.43064D-01      8.79443D+07 NIST class=A

1-5873.93740296 2s(2).2p(6).3s(2).3p(6).3d(10).4s(2).4p(6).4d(10).5s(2).6s_2S
3-5873.90217166 2s(2).2p(6).3s(2).3p(6).3d(10).4s(2).4p(6).4d(10).5s(2).6p_2P
7732.38 CM-1 12932.64 ANGSA(VAC) 12931.30 ANGSA(AIR)
E1 S = 7.36341D+01 GF = 1.72948D+00 AKI = 1.72435D+07 dT = 0.00500
      7.40044D+01      1.73818D+00      1.73302D+07 NIST class=AA

1-5873.93740296 2s(2).2p(6).3s(2).3p(6).3d(10).4s(2).4p(6).4d(10).5s(2).6s_2S
3-5873.87107450 2s(2).2p(6).3s(2).3p(6).3d(10).4s(2).4p(6).4d(10).5s(2).7p_2P
14557.41 CM-1 6869.35 ANGSA(VAC) 6868.64 ANGSA(AIR)
E1 S = 1.30939D+00 GF = 5.78998D-02 AKI = 2.04610D+06 dT = 0.01201
      1.32531D+00      5.86038D-02      2.07098D+06 NIST class=A+

3-5874.03780394 2s(2).2p(6).3s(2).3p(6).3d(10).4s(2).4p(6).4d(10).5s(2).5p_2P
1-5873.88326461 2s(2).2p(6).3s(2).3p(6).3d(10).4s(2).4p(6).4d(10).5s(2).7s_2S
33917.46 CM-1 2948.33 ANGSA(VAC) 2948.02 ANGSA(AIR)
E1 S = 5.60932D-01 GF = 5.77907D-02 AKI = 2.21726D+07 dT = 0.01252
      5.68042D-01      5.85232D-02      2.24536D+07 NIST class=A+

3-5873.90217166 2s(2).2p(6).3s(2).3p(6).3d(10).4s(2).4p(6).4d(10).5s(2).6p_2P
1-5873.88326461 2s(2).2p(6).3s(2).3p(6).3d(10).4s(2).4p(6).4d(10).5s(2).7s_2S
4149.62 CM-1 24098.61 ANGSA(VAC) 24096.12 ANGSA(AIR)
E1 S = 8.87672D+01 GF = 1.11888D+00 AKI = 6.42559D+06 dT = 0.01503
      8.74331D+01      1.10207D+00      6.32902D+06 NIST class=A+

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Summary and conclusions

- ▶ layer-by-layer approach has limitations as the valence orbitals fixed from DF
- ▶ NO basis accounts for the valence orbital contraction due to interaction with the core
- ▶ sizeable effects on hyperfine structure and transition rates
- ▶ effect largest for neutral systems
- ▶ transformation to NOs can (should?) be done based on SMR expansion (good since these expansions are small)
- ▶ systematic analysis of the effect of NOs can be done using perturbation theory, Gediminas has preliminary results

Thank you for listening!