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Laser Spectroscopy of Stable and Radioactive Negative Ions

DAG HANSTORP

CompAS 2025, Lund, Sweden

12-14 June 2025

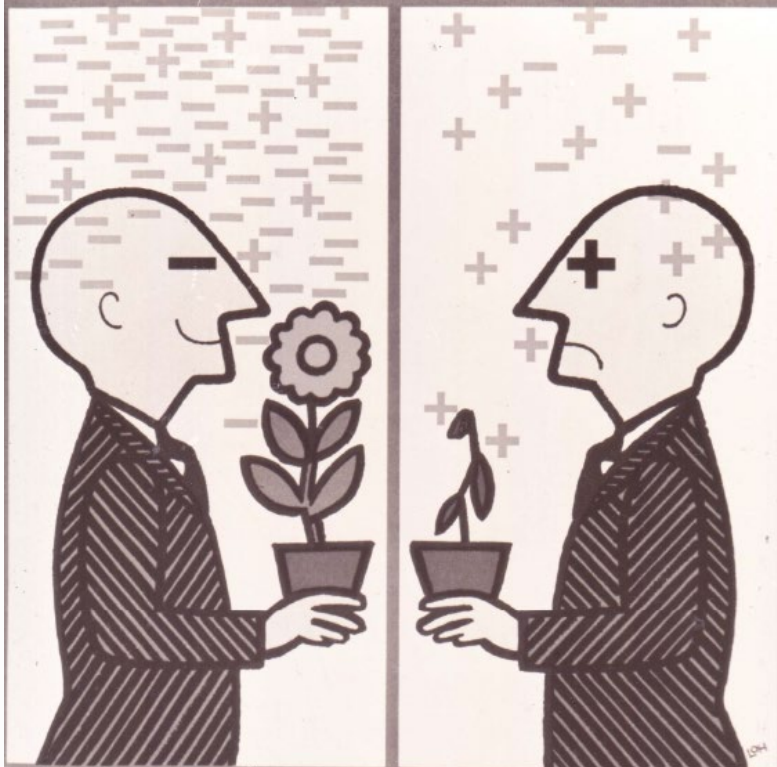
new scientist

14 June 1973

Vol 58 No 850

Weekly 15p

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Are negative ions good for you?

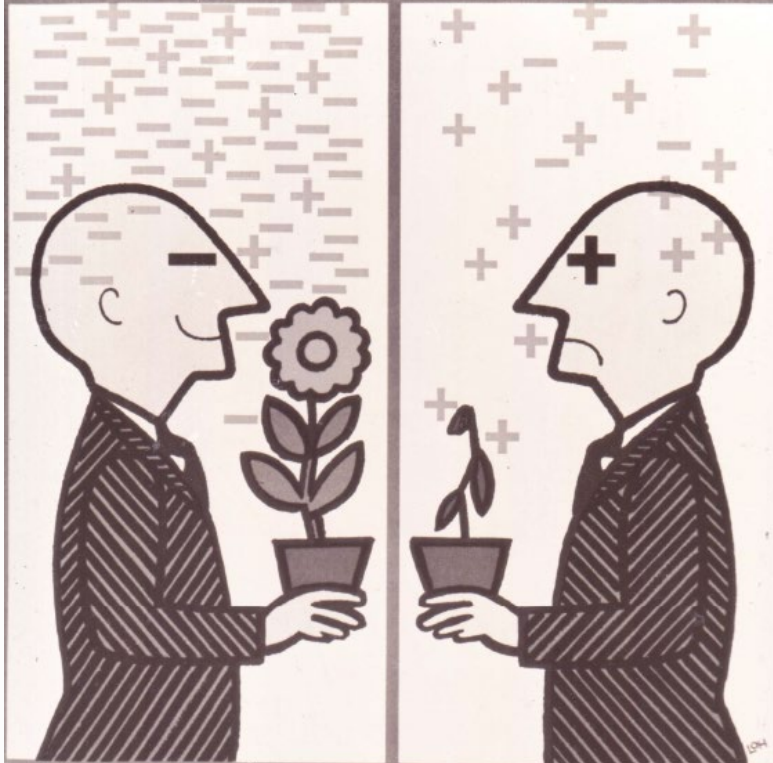
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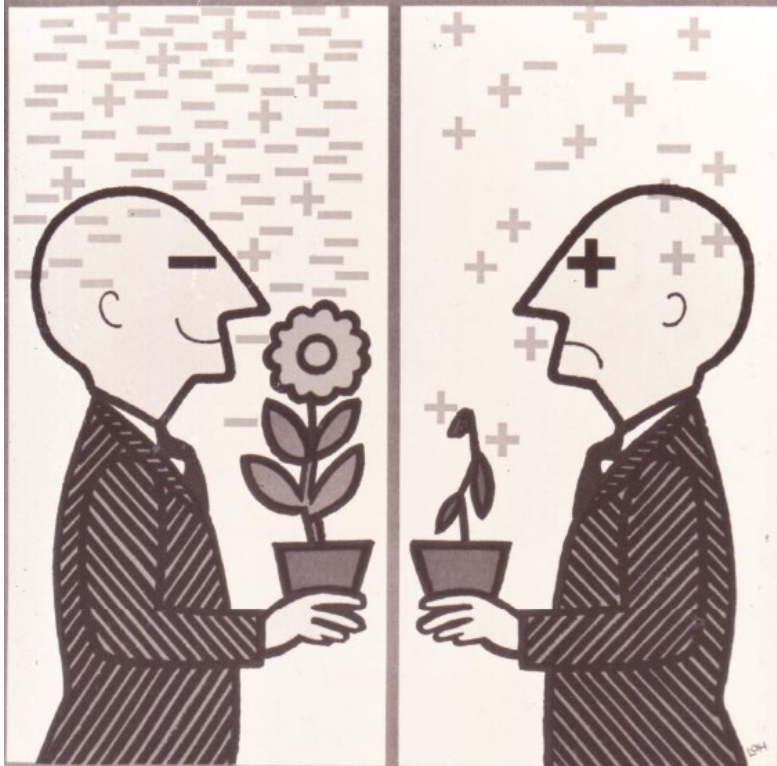
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USA (by air) 90 cents/
BF 25/FF 3/DM 2.80/
hfi 1.75/ukr 3.00/



Are negative ions good for you?



Eric Berg



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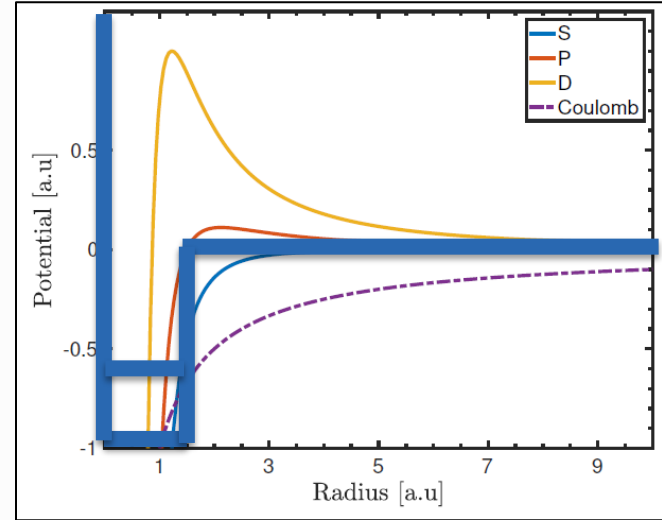
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Negative ions



$$H = H_e + H_{ee} + V(r_i) = -\frac{\hbar^2}{2m_e} \sum_i \Delta_i + \sum_{i,j,i \neq j} \frac{e^2}{4\pi\epsilon_0 r_{ij}} - \frac{Ze^2}{4\pi\epsilon_0 r_i}$$



For an atom:

$$V_C(r) = -\frac{e^2}{4\pi\epsilon_0 r}$$

For a negative ion:

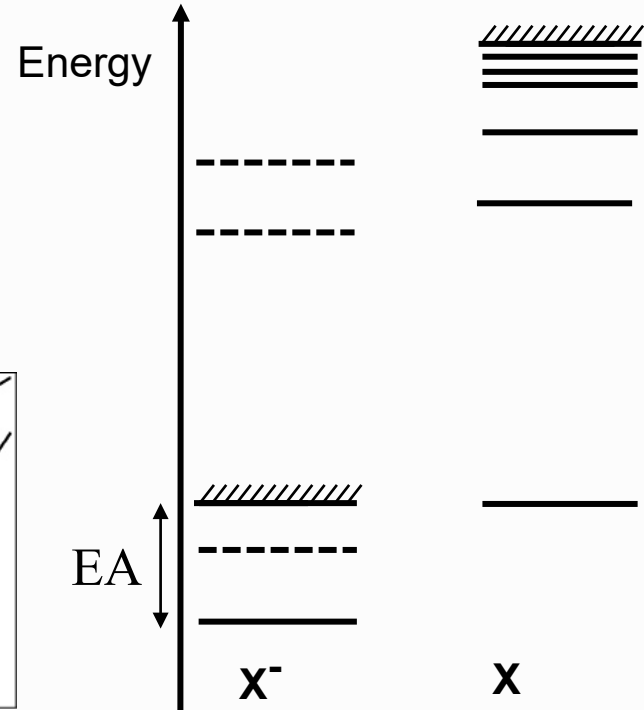
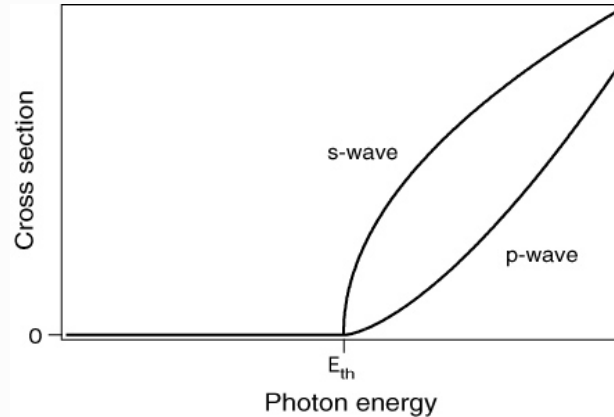
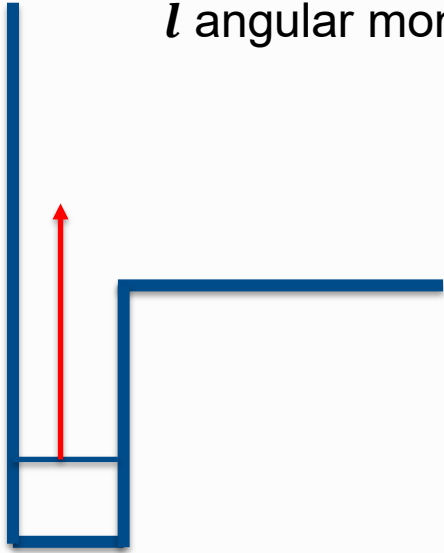
$$V(r) = \frac{l(l+1)}{2r^2} - \frac{\alpha_D}{2r^4}$$

Photodetachment: $hf + A^- \rightarrow A + e^-$

Wigner threshold law:

$$\sigma(E) = k (E_\gamma - E_{th})^{l+\frac{1}{2}}$$

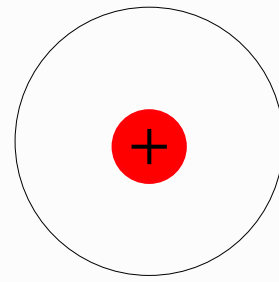
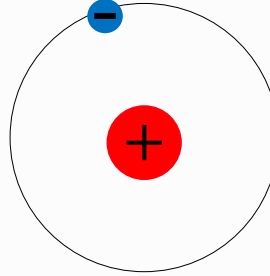
l angular momentum of emitted electron



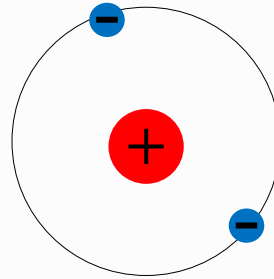
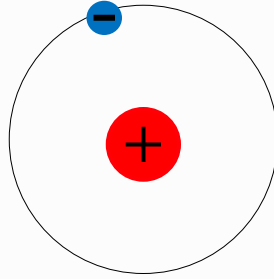
Why study negative ions?

- Benchmark for electron correlation theory
- Model system in physics
- Appears in many environments and used applications

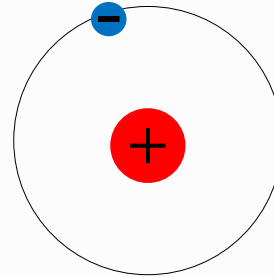
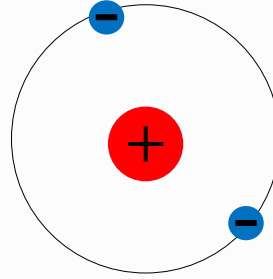
Producing a positive hydrogen ion:



Producing a negative hydrogen ion:



Study negative ions:



The electron affinity of Astatine

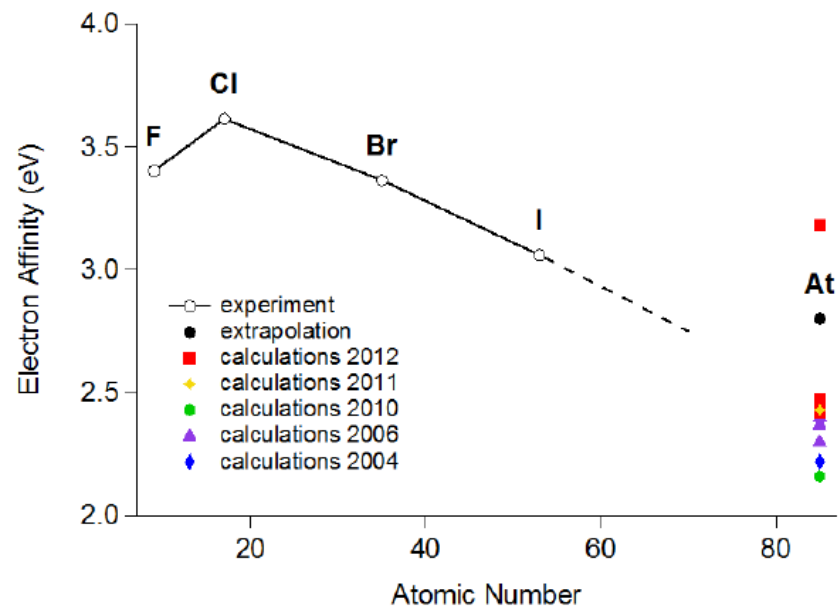
Periodic Table of Elements

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
1 H Hydrogen 1.00794	2 He Helium 4.002602																
3 Li Lithium 6.941	4 Be Beryllium 9.012182																
5 Na Sodium 22.98976928	6 Mg Magnesium 24.304																
7 K Potassium 39.0983	8 Ca Calcium 40.078	9 Sc Scandium 44.955912	10 Ti Titanium 47.88	11 V Vanadium 50.9415	12 Cr Chromium 51.9961	13 Mn Manganese 54.938045	14 Fe Iron 55.845	15 Co Cobalt 58.933195	16 Ni Nickel 58.6934	17 Cu Copper 63.546	18 Zn Zinc 65.38	19 Ga Gallium 69.723	20 Ge Germanium 72.64	21 As Arsenic 74.9216	22 Se Selenium 78.96	23 Br Bromine 79.904	24 Kr Krypton 83.798
19 Rb Rubidium 85.4678	20 Sr Strontium 87.62	21 Y Yttrium 88.90585	22 Zr Zirconium 91.224	23 Nb Niobium 92.90638	24 Mo Molybdenum 95.94	25 Tc Technetium 97.9072	26 Ru Ruthenium 101.07	27 Rh Rhodium 102.9055	28 Pd Palladium 106.42	29 Ag Silver 107.8682	30 Cd Cadmium 112.411	31 In Indium 114.818	32 Sn Tin 118.710	33 Sb Antimony 121.757	34 Te Tellurium 127.6	35 I Iodine 126.90547	36 Xe Xenon 131.29
37 Cs Cesium 132.90545196	38 Ba Barium 137.327	39 La Lanthanum 138.90547	40 Ce Cerium 140.12	41 Pr Praseodymium 140.90768	42 Nd Neodymium 144.242	43 Pm Promethium (144.9127)	44 Sm Samarium 150.36	45 Eu Europium 151.964	46 Gd Gadolinium 157.25	47 Tb Terbium 158.92535	48 Dy Dysprosium 162.5003	49 Ho Holmium 164.93032	50 Er Erbium 167.259	51 Tm Thulium 168.93042	52 Yb Ytterbium 173.054	53 Lu Lutetium 174.967	54 Hf Hafnium 178.49
55 Rb Rubidium 85.4678	56 Sr Strontium 87.62	57-71 Lanthanides	72 Hf Hafnium 178.49	73 Ta Tantalum 180.94788	74 W Tungsten 183.84	75 Re Rhenium 186.207	76 Os Osmium 190.23	77 Ir Iridium 192.222	78 Pt Platinum 195.084	79 Au Gold 196.96657	80 Hg Mercury 200.59	81 Tl Thallium 204.3833	82 Pb Lead 207.2	83 Bi Bismuth 208.9804	84 Po Polonium 209	85 At Astatine (210)	86 Rn Radon 222
87 Fr Francium (223)	88 Ra Radium (226)	89-103 Actinides	104 Rf Rutherfordium (261)	105 Db Dubnium (262)	106 Sg Seaborgium (266)	107 Bh Bohrium (264)	108 Hs Hassium (277)	109 Mt Meitnerium (268)	110 Ds Darmstadtium (271)	111 Rg Roentgenium (272)	112 Uub Ununbium (285)	113 Uut Ununtrium (284)	114 Uuq Ununquadium (289)	115 Uup Ununpentium (288)	116 Uuh Ununhexium (292)	117 Uus Ununseptium (294)	118 Uuo Ununoctium (294)

For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.

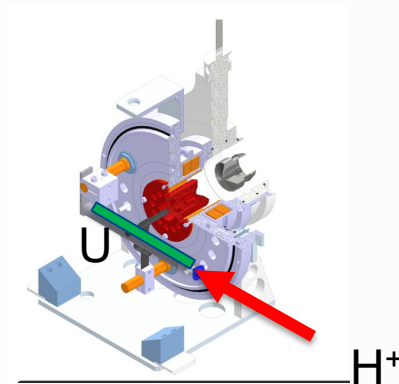
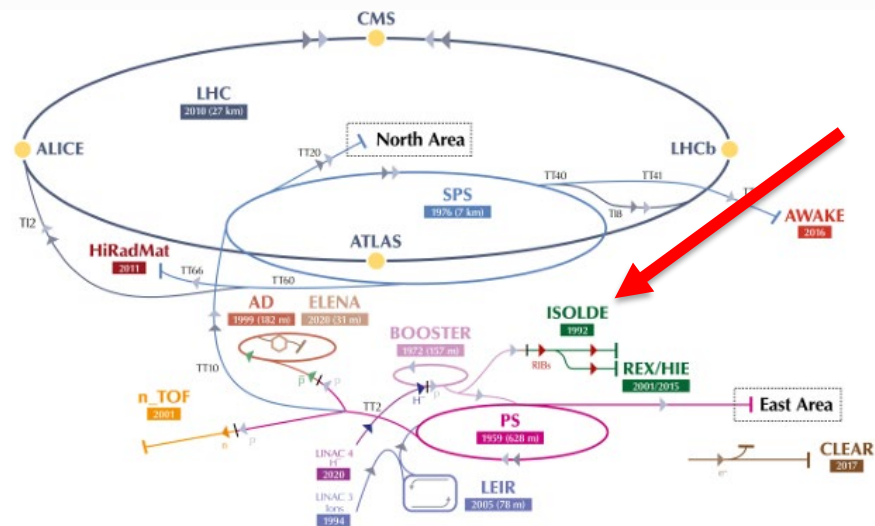
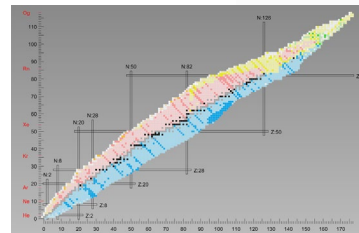
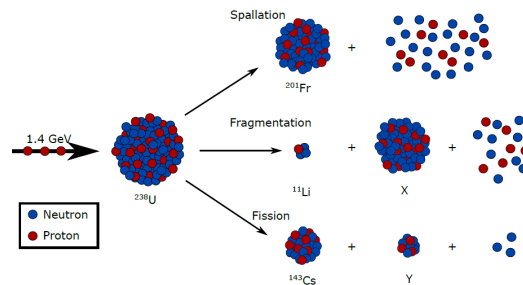
Design and Interface Copyright © 1997 Michael Dayah (michael@dayah.com). <http://www.ptable.com/>

- Least abundant element on earth
- 70 mg in the crust of the earth (1 atom per 100 kg mass)
- Decays through α -decay
- Used in cancer treatment Targeted Alfa Therapy (TAT) (suitable lifetime and energy, non-toxic, non-radioactive daughters)
- Small knowledge about its chemical and physical properties



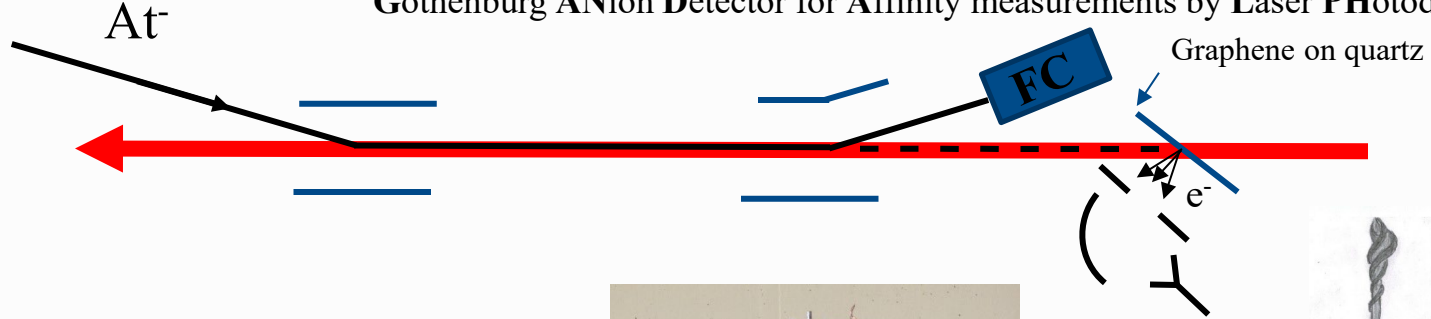


ISOLDE



GANDALPH

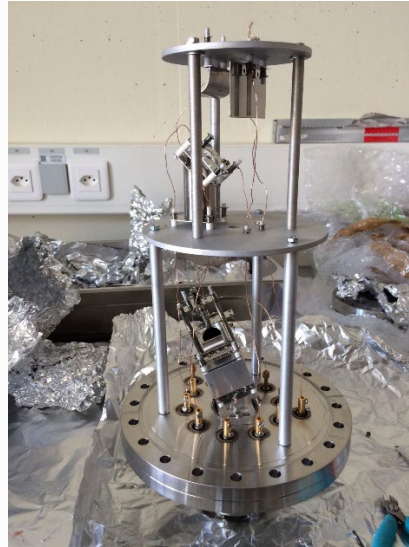
Gothenburg ANion Detector for Affinity measurements by Laser PHotodetachment



For each laserpuls:

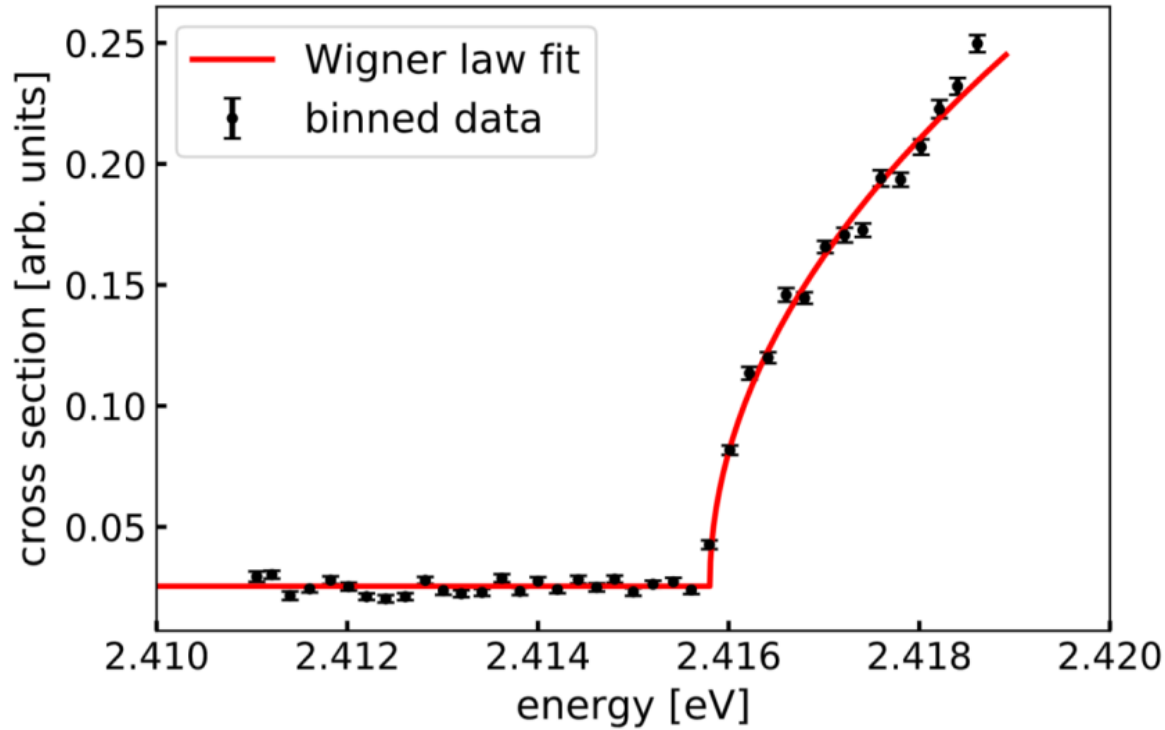
Signal:
0.01 atom

Background:
 10^{14} photons



Drawing:
Annie
Ringvall
-Moberg

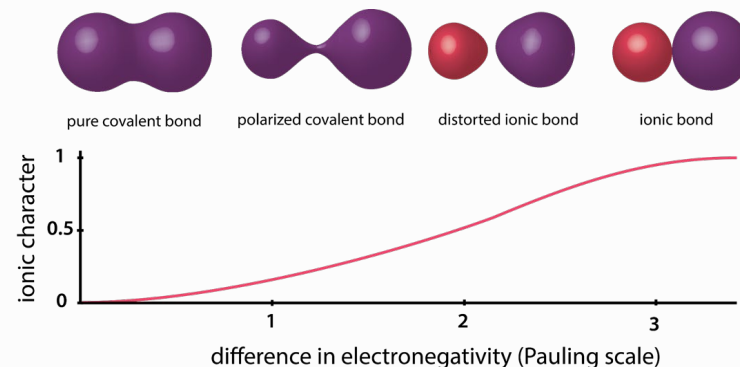
RESULTS



Experiment: EA= 2.415 78 (5) eV

Method	Author	Year	EA
Experiment			2.41578(5)
CBS-DC-CCSDT(Q)+Breit+QED			2.414(16)
MCDHF+SE corr.	Chang et al	2010	2.38(2)
MCDHF	Zhao et al	2012	2.416
DC-CCSD(T)+Breit+QED	Broschevsky et al	2015	2.412
MCDHF+Extrap.+Breit+QED	Si and Fischer	2018	2.3729(46)
CBS-DC-CCSD(T)+Gaunt+QED	Finney and Peterson	2019	2.423(13)

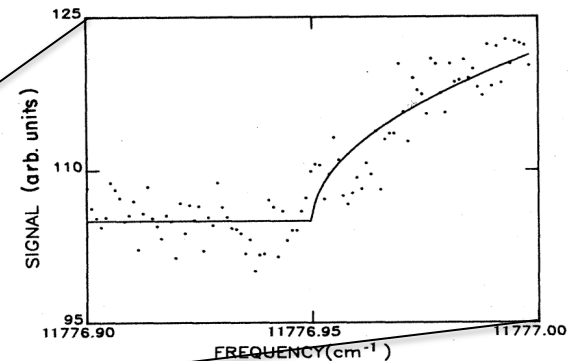
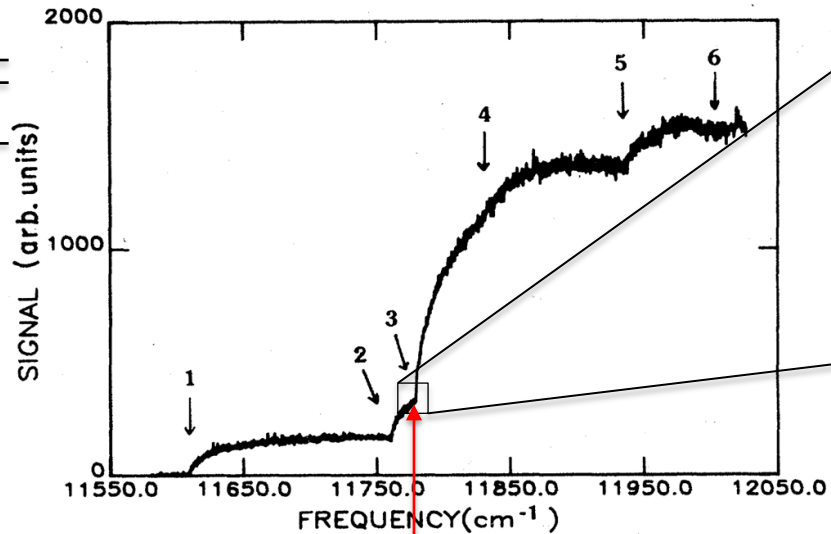
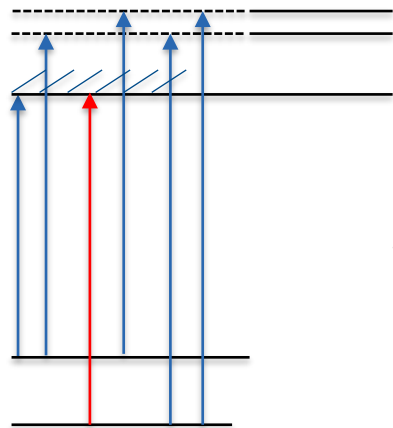
Property	Definition	Value
Electron affinity	EA	2.415 78(5) eV
Ionization energy	IE	9.317 51(8) eV ²⁰
Electronegativity	$\chi_M = \frac{IP+EA}{2}$	5.866 65 eV
Hardness	$\eta = IE - EA$	6.901 72(13) eV
Softness	$S = \frac{1}{\eta}$	0.144 89(2) eV ⁻¹
Electrophilicity	$\omega = \frac{\chi^2}{2\eta}$	2.493 41(8) eV



The At-H molecule should be called **astatine hydride** (not hydrogen astatide)

Leimbach *et al.* Nature Communications, 11 (2020) 3824

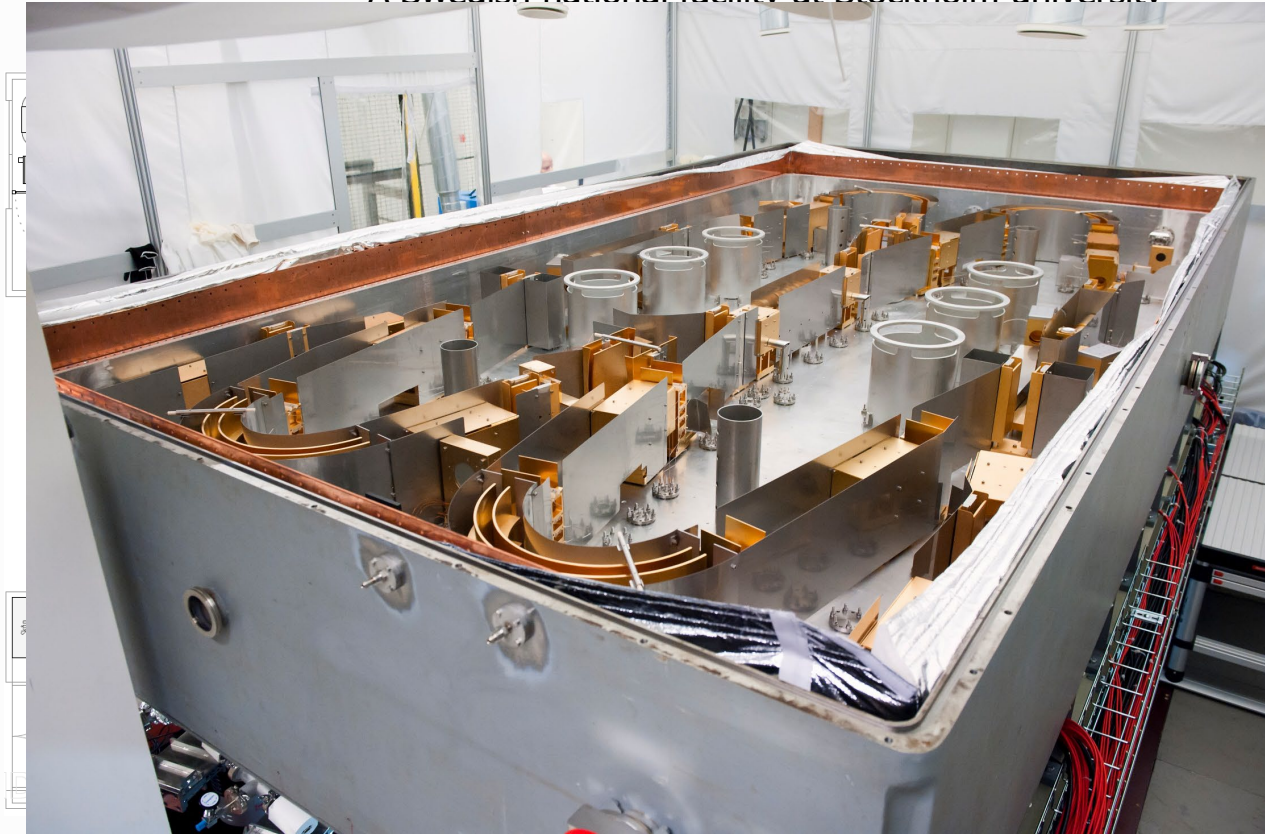
Precision measurement of the EA of O^-



Neumark, D. M., et al. Physical Review A 32.3 (1985): 1890

DESIREE – Double Electrostatic Ion Ring Experiment

A Swedish national facility at Stockholm university

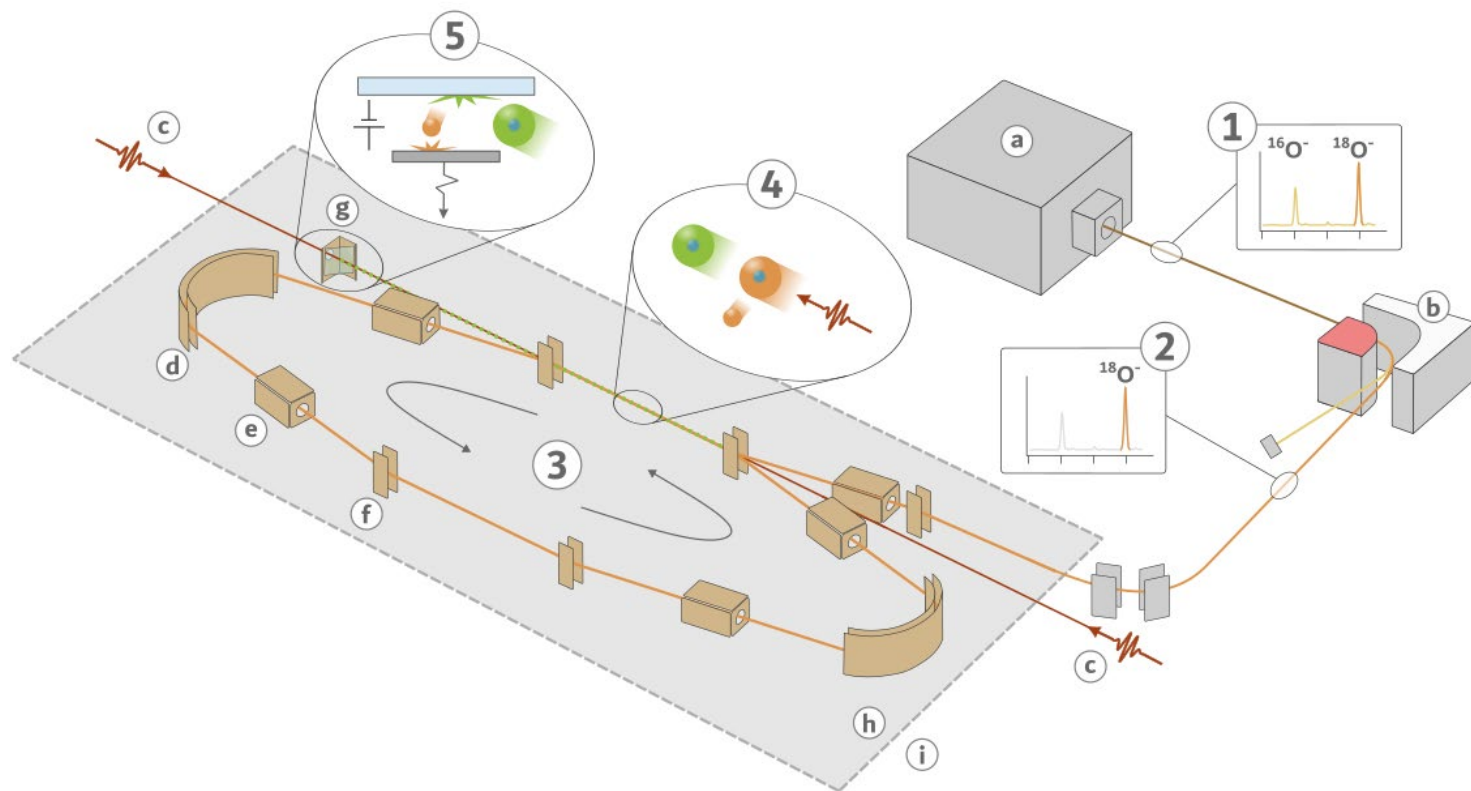


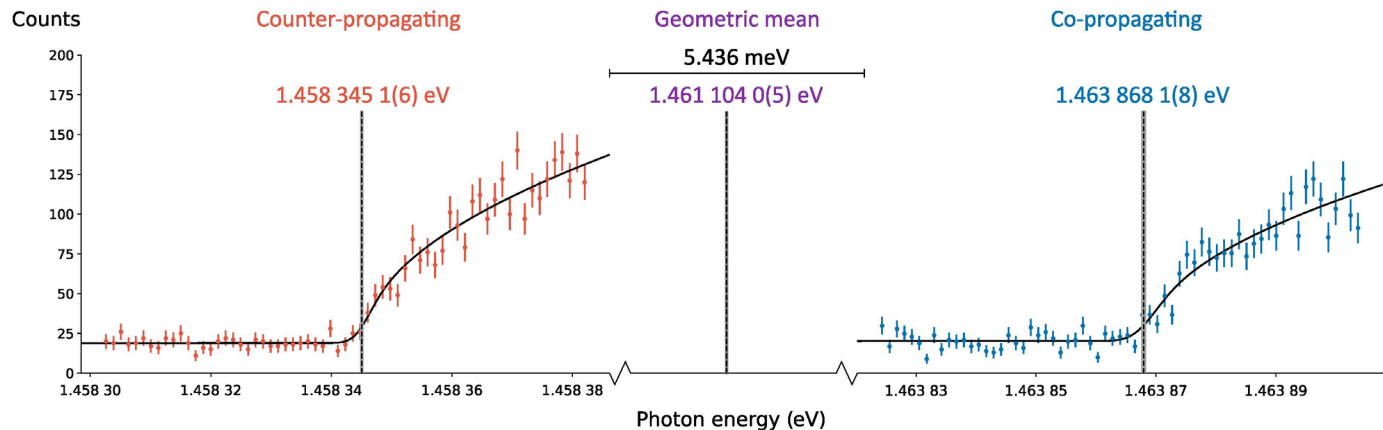
nds amu

- keV

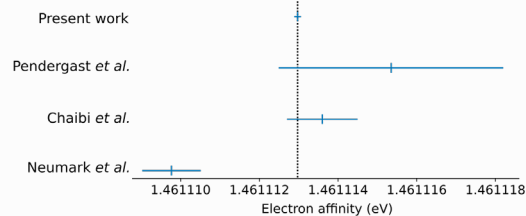
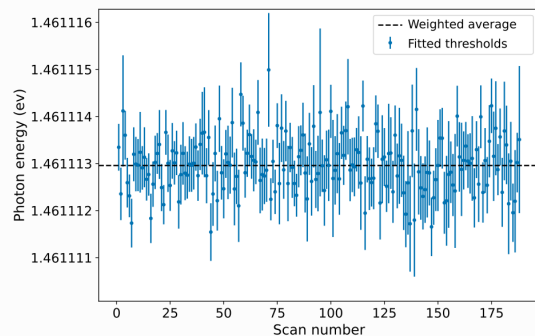
years
MEURO

→ *Physics at conditions of the interstellar media*





$$\sqrt{E_{EA}^p E_{EA}^a} = \sqrt{\frac{1 + v/c}{\sqrt{1 - v^2/c^2}} E_{EA} \frac{1 - v/c}{\sqrt{1 - v^2/c^2}} E_{EA}} = \sqrt{E_{EA} E_{EA}} = E_{EA}$$



Final result: $E_{EA} = 1.461\,112\,972\,(87)\,\text{eV}$

Kristiansson, *et al.* Nature Communications **13** (2022) 5906

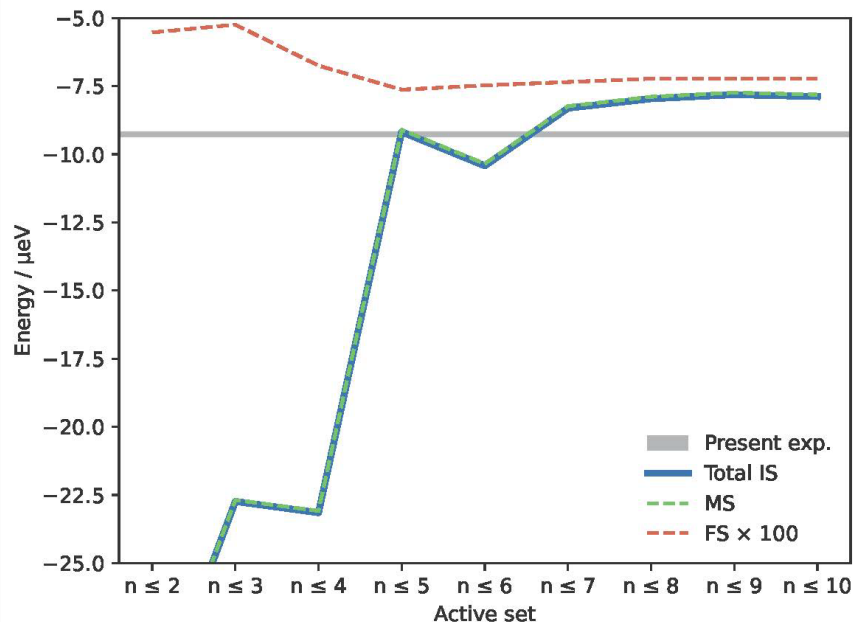
¹⁸⁻¹⁶O isotope shift

$$E_{EA}(^{16}\text{O}) = 1.461\,112\,972\,(87)\,\text{eV}$$

$$E_{EA}(^{18}\text{O}) = 1.461\,103\,706\,(67)\,\text{eV}$$

$$IS = -0.000\,009\,267(11)\,\text{eV}$$

Multi-configurational Dirac-Hartree-Fock (MCDHF) and relativistic configuration-interaction (RCI) methods, as implemented in the current development version of GRASP2018 (Jon Grumer)



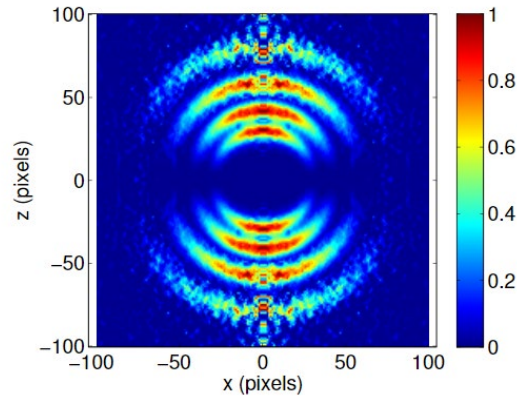
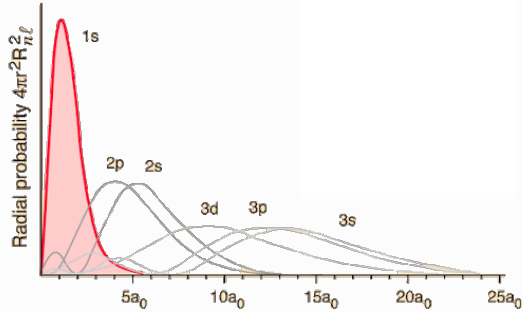
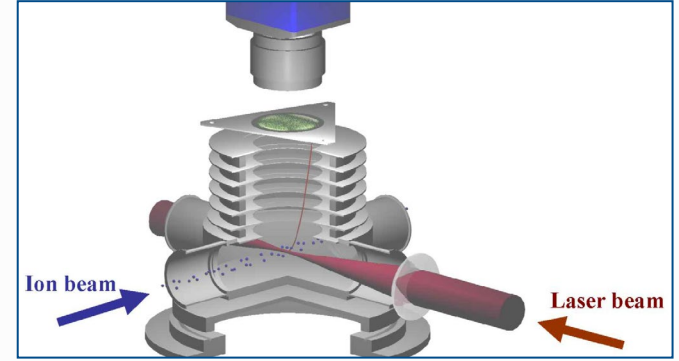
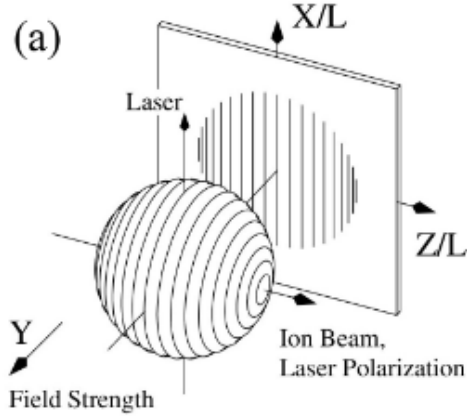
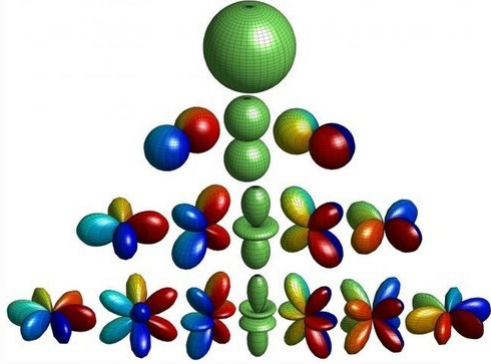
New experimental value -9.27(0.09) μeV

Previous IS experimental value: -9.2(2.2) μeV
C. Blondel, Physical Review A 64, 052504 (2001).

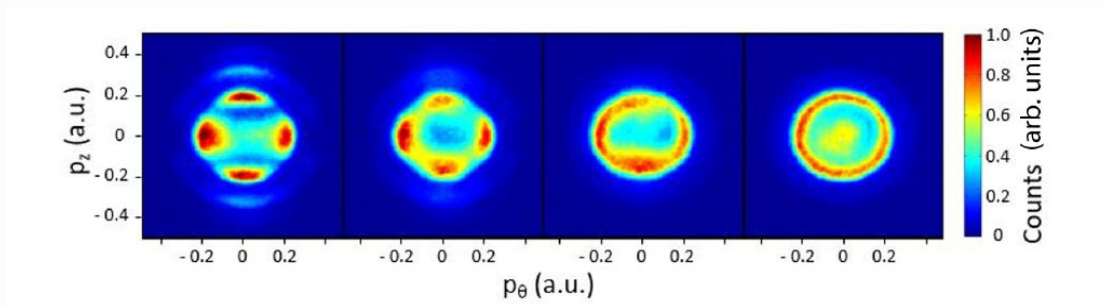
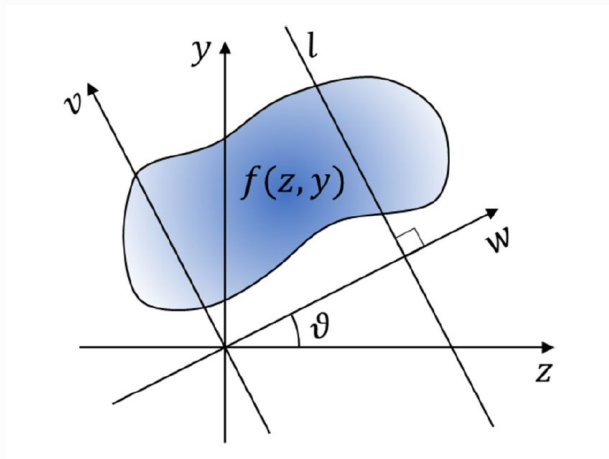
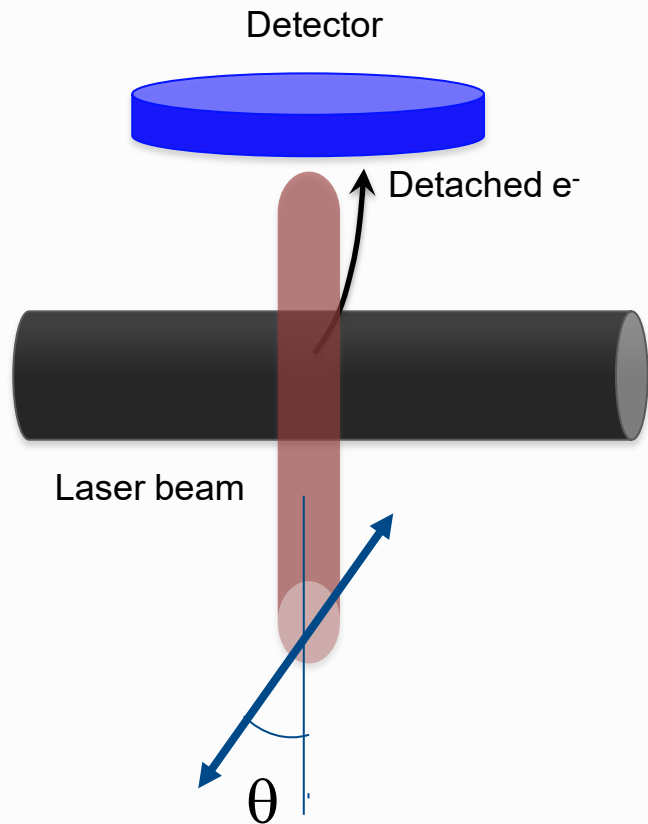
New theoretical value: -7.884 μeV

Previous Theoretical value: -7.104 μeV
Godefroid and C. F. Fischer,
Phys. Rev. A 60, R2637 (1999).

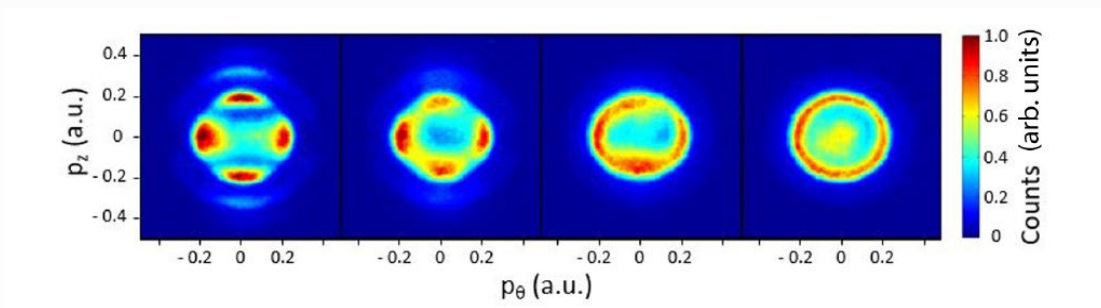
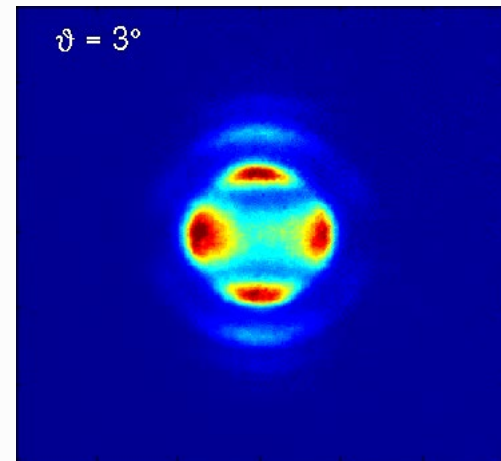
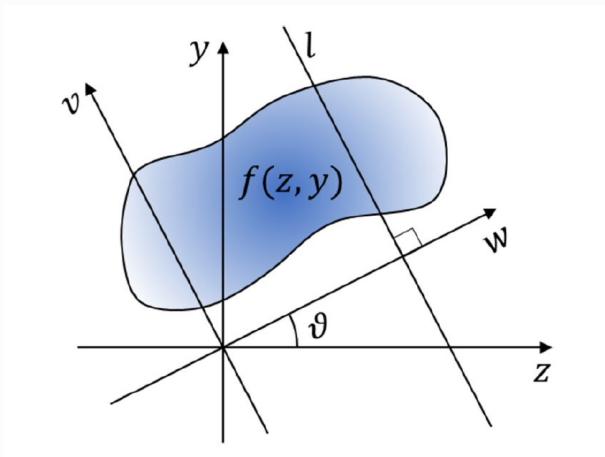
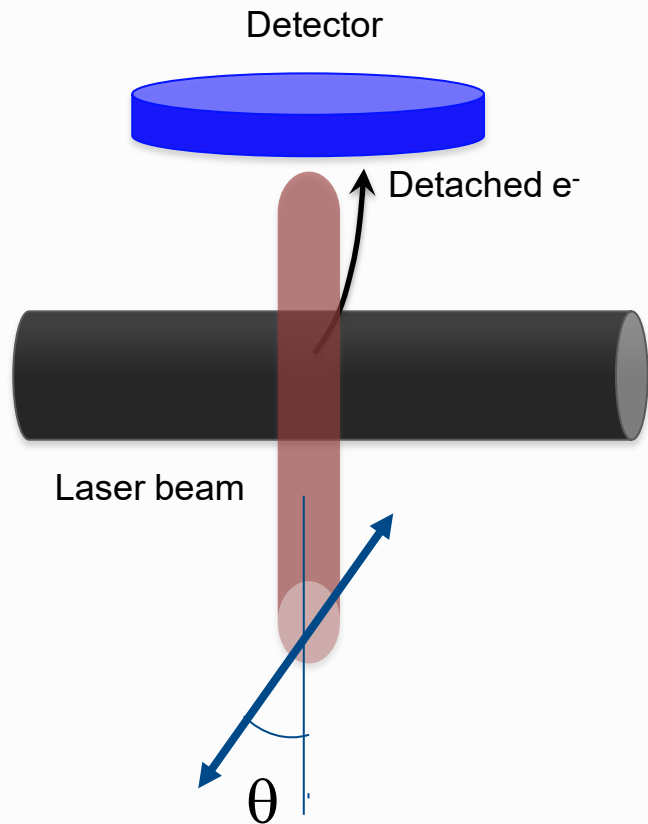
Probing wavefunctions: Velocity map imaging (VMI)



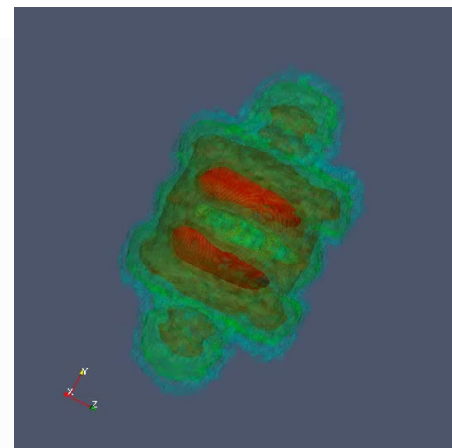
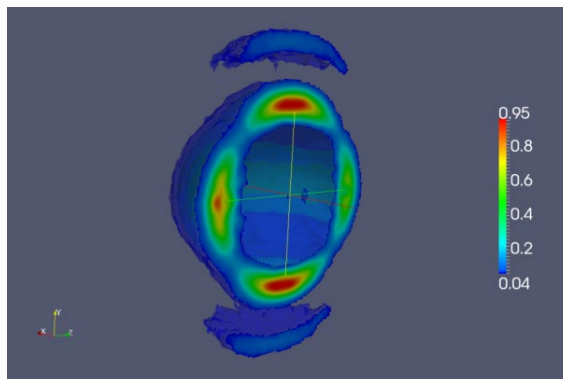
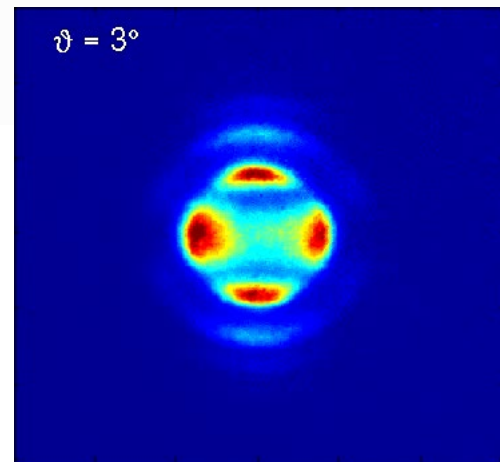
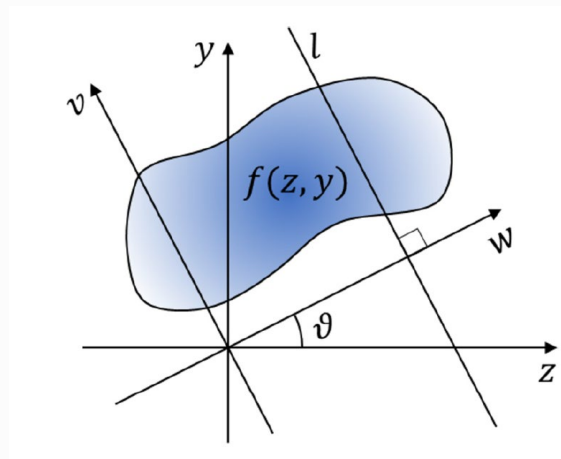
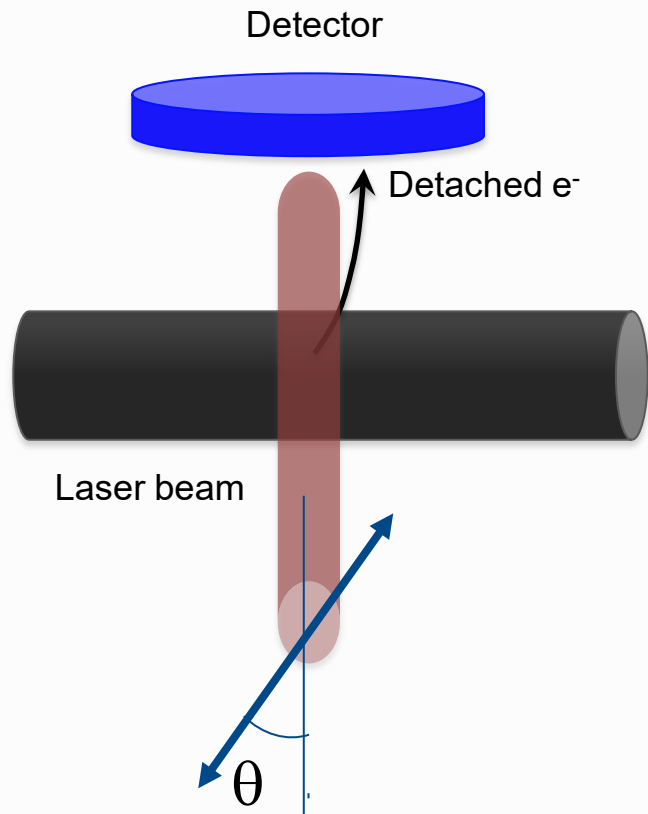
Optical tomography



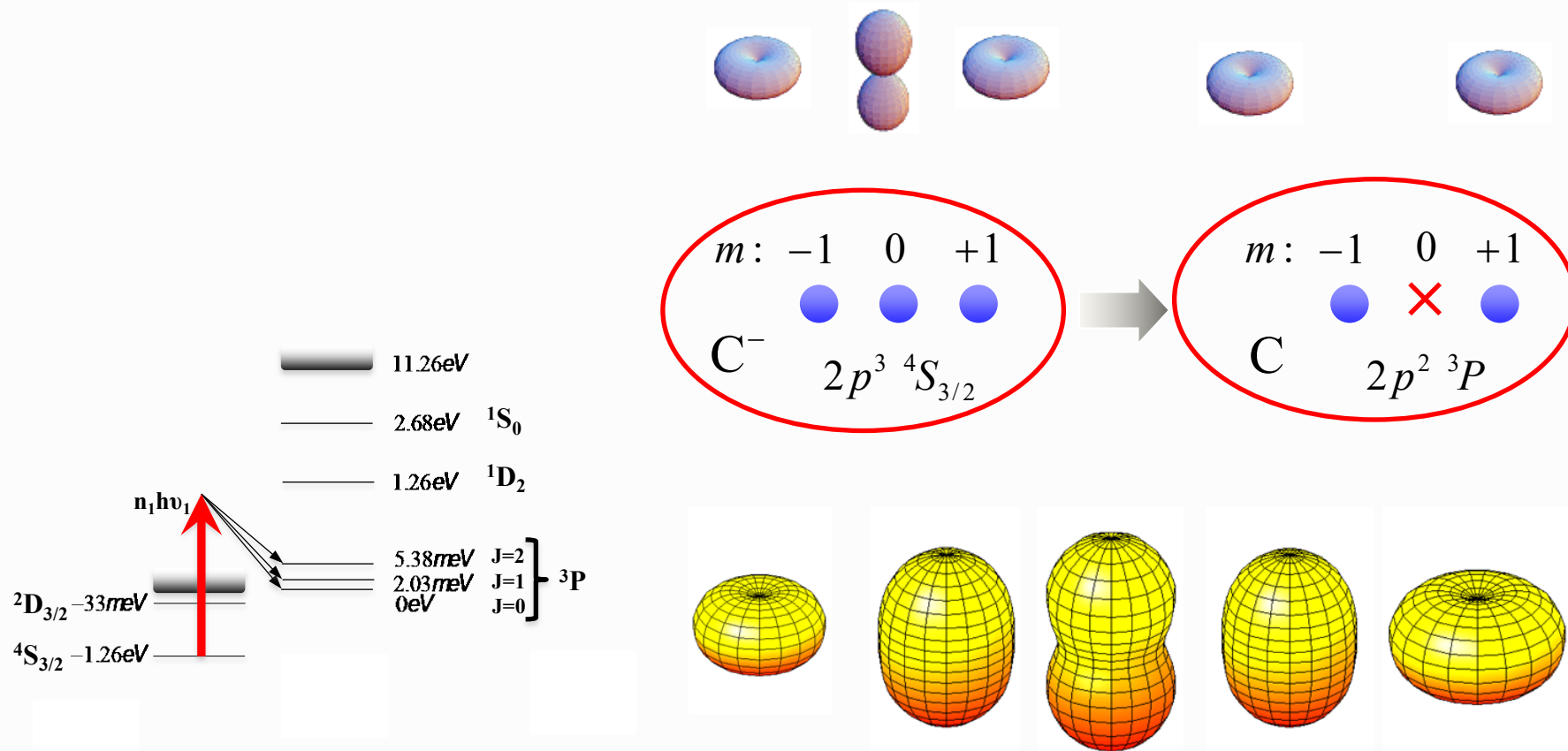
Optical tomography

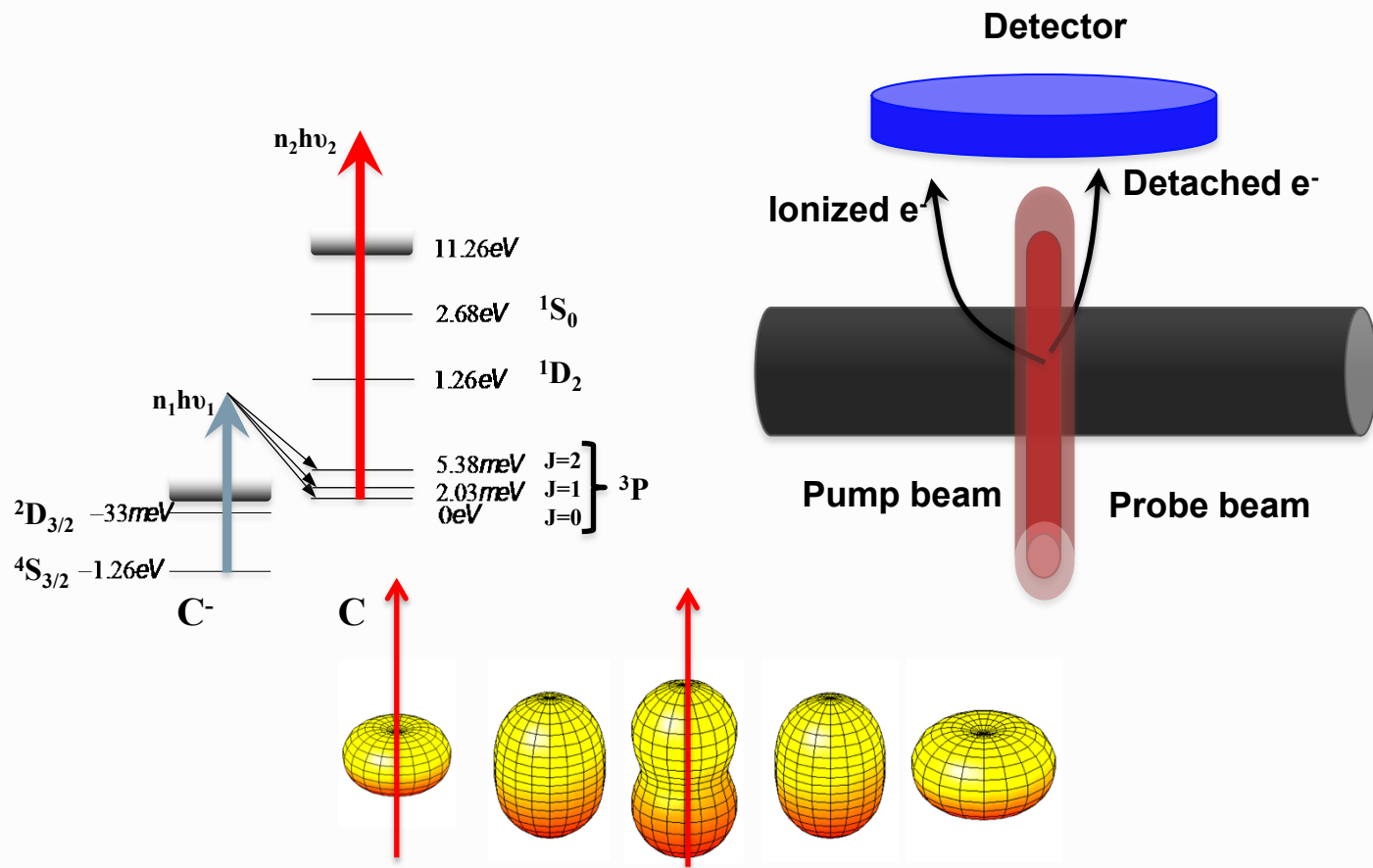


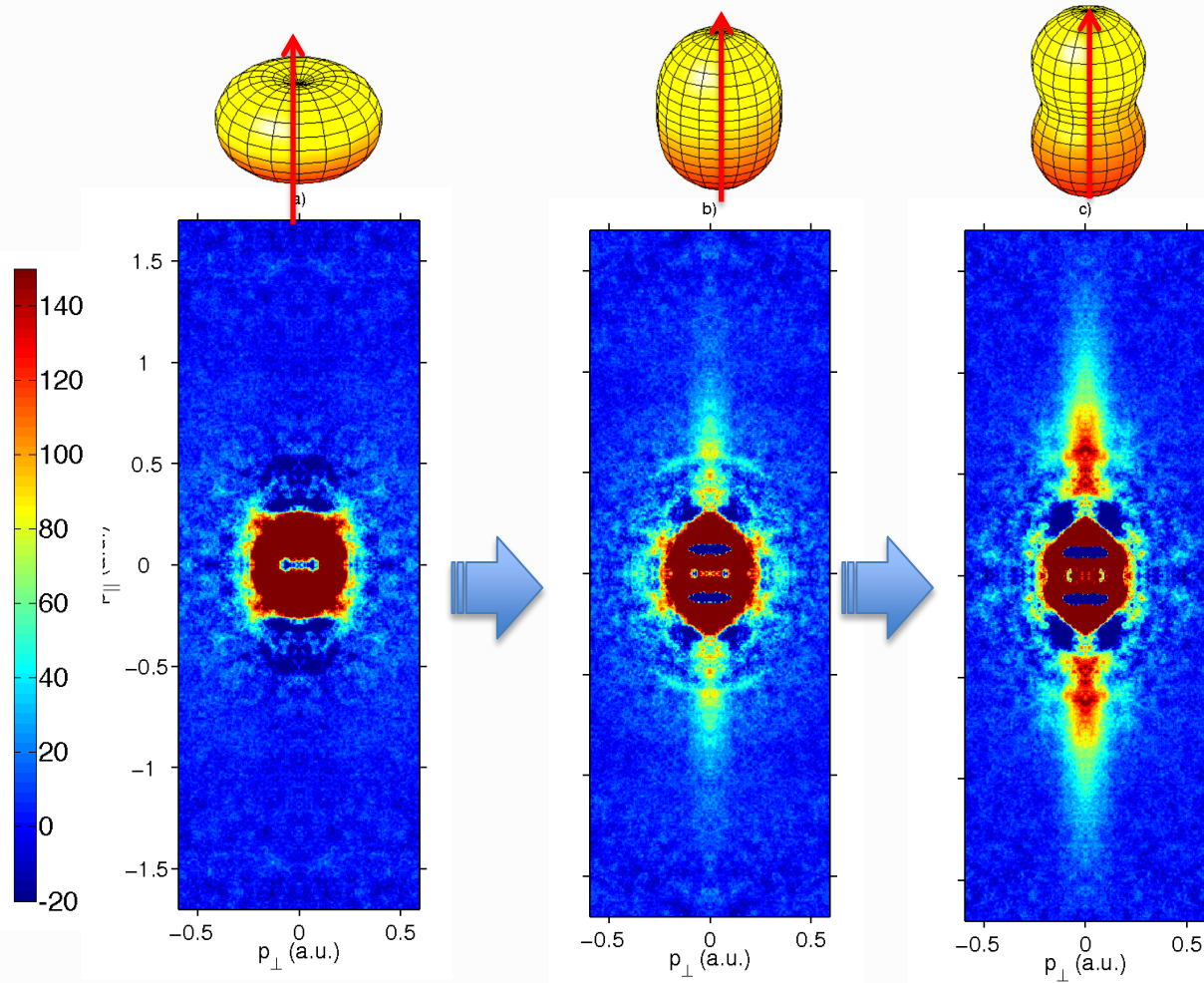
Optical tomography

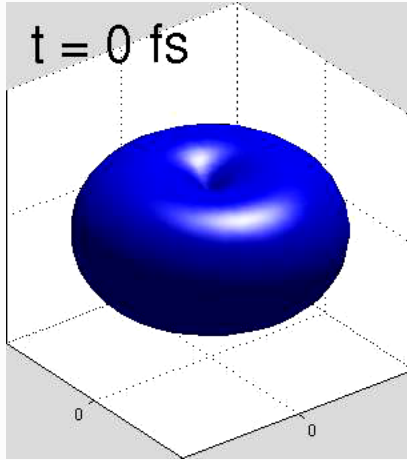


Photodetachment in a strong field:

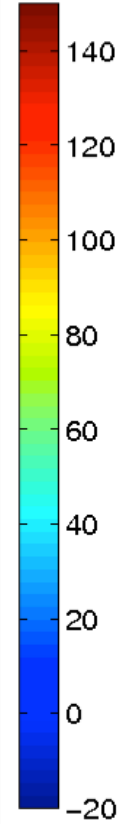
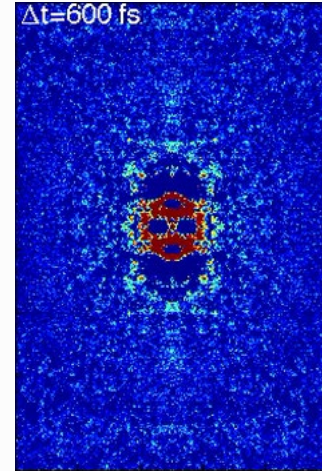








Laser Polarization



Hultgren, Mikael Eklund, Dag Hanstorp, and Igor Yu. Kiyan,
Phys. Rev. A 87, (2013) 031404.

Detection of photoemitted electrons



Energy → structure of the ion

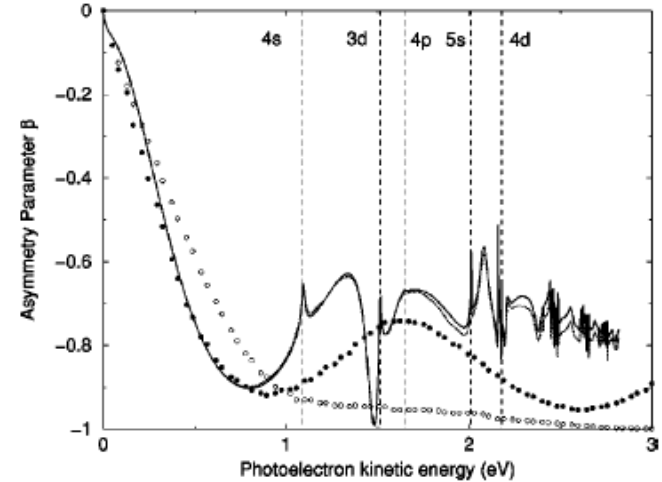
Angular distribution → relative phases of emitted electron waves the symmetry of the negative ion

$$\frac{d\sigma}{d\Omega} = \left[\frac{\sigma}{4\pi} \right] [1 + \beta P_2(\cos\theta)]$$

$$P_2(\cos\theta) = (3\cos^2\theta - 1)/2,$$

$$-1 < \beta < 2$$

$$\begin{aligned} \beta = -1 & \quad \sin^2(\theta) \\ \beta = 0 & \quad \text{isotropic} \\ \beta = 2 & \quad \cos^2(\theta) \end{aligned}$$



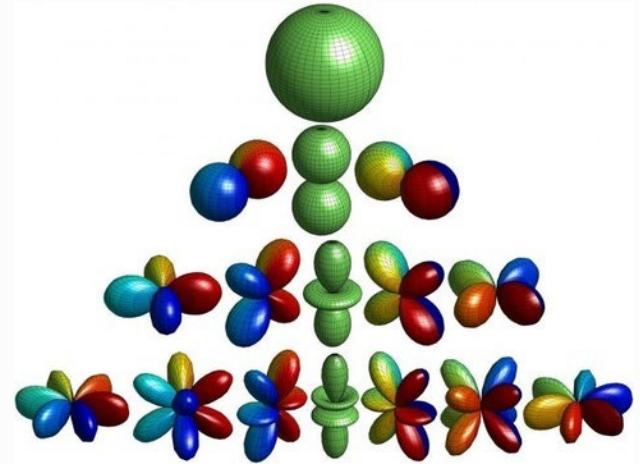
Photodetachment of $\text{Na}^- + h\nu \rightarrow \text{Na}(3p) + e^-$

Liu and Starace PRA, 59 (1999) 3643

$$X^-(ns) + h\nu \rightarrow X + e^- (\varepsilon p) \quad (\beta = 2)$$

$$X^-(np) + h\nu \rightarrow X + e^- (\varepsilon s) \quad (\beta = 0)$$

$$\rightarrow X + e^- (\varepsilon d) \quad (\beta = +)$$



Two waves $\rightarrow$ Interference

Cooper and Zare, J. Chem. Phys. 48 (1968) 942:

$$\beta = \frac{l(l-1)R_{l-1}^2 + (l+1)(l+2)R_{l+1}^2 - 6l(l+1)R_{l+1}R_{l-1}\cos(\delta_{l+1} - \delta_{l-1})}{(2l+1)[lR_{l-1}^2 + (l+1)R_{l+1}^2]},$$

$$\beta = \frac{l(l-1)R_{l-1}^2 + (l+1)(l+2)R_{l+1}^2 - 6l(l+1)R_{l+1}R_{l-1}\cos(\delta_{l+1} - \delta_{l-1})}{(2l+1)[lR_{l-1}^2 + (l+1)R_{l+1}^2]},$$

Photodetachment of P⁻

Windelius *et al.* Rev. A. **103** (2021) 033108.

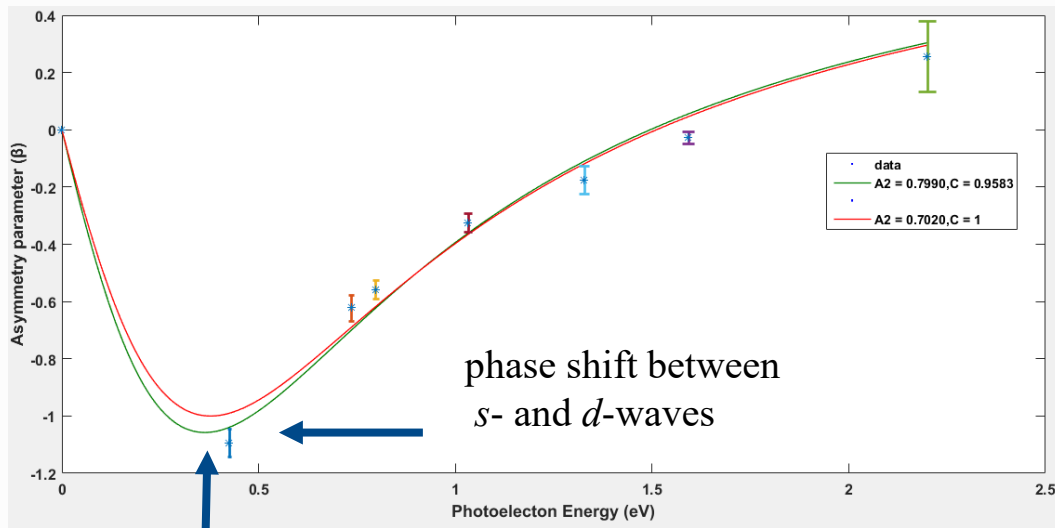
Wigner law:

$$R_{l\pm 1} \sim k^{l\pm 1}$$

$$R_2/R_0 = \frac{k^{l+1}}{k^{l-1}} = k^2 = A_2\varepsilon$$

$$c = (\delta_{l+1} - \delta_{l-1})$$

$$\beta = 2A_2\varepsilon(A_2\varepsilon - 2c)/(1 + 2A_2^2\varepsilon^2)$$



Amplitude of
s- and *d*-waves are equal

Photoelectron angular distributions for states of any mixed character: An experiment-friendly model for atomic, molecular, and cluster anions

Dmitry Khuseynov, Christopher C. Blackstone, Lori M. Culberson,^{a)} and Andrei Sanov^{b)}

Department of Chemistry and Biochemistry, University of Arizona, Tucson, Arizona 85721, USA

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We present a model for laboratory-frame photoelectron angular distributions in direct photodetachment from (in principle) any molecular orbital using linearly polarized light. A transparent mathematical approach is used to generalize the Cooper-Zare central-potential model to anionic states of any mixed character. In the limit of atomic-anion photodetachment, the model reproduces the Cooper-Zare formula. In the case of an initial orbital described as a superposition of s and p -type functions, the model yields the previously obtained s - p mixing formula. The formalism is further advanced using the Hanstorp approximation, whereas the relative scaling of the partial-wave cross-sections is assumed to follow the Wigner threshold law. The resulting model describes the energy dependence of photoelectron anisotropy for any atomic, molecular, or cluster anions, usually without requiring a direct calculation of the transition dipole matrix elements. As a benchmark case, we apply the p - d variant of the model to the experimental results for NO^- photodetachment and show that the observed anisotropy trend is described well using physically meaningful values of the model parameters. Overall, the presented formalism delivers insight into the photodetachment process and affords a new quantitative strategy for analyzing the photoelectron angular distributions and characterizing mixed-character molecular orbitals using photoelectron imaging spectroscopy of negative ions. © 2014 AIP Publishing LLC. [<http://dx.doi.org/10.1063/1.4896241>]

C. Examples of binary mixing curves

As discussed previously,^{3,24} the s - p mixing equation with the Hanstorp coefficients, Eq. (33), can be rearranged as follows:

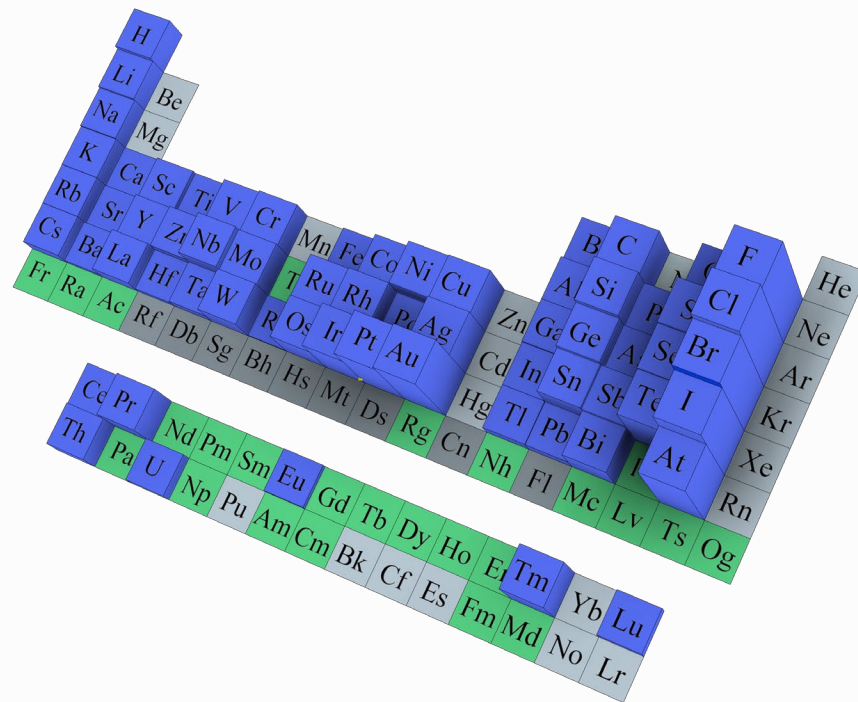
$$\beta_{sp}(\varepsilon) = \frac{2Z_1(A_1\varepsilon) + 2(A_1\varepsilon)^2 - 4(A_1\varepsilon)\cos\delta_{2,0}}{Z_1(A_1\varepsilon) + 1 + 2(A_1\varepsilon)^2}, \quad (40)$$

What do we know about negative ions?

Many EAs determined

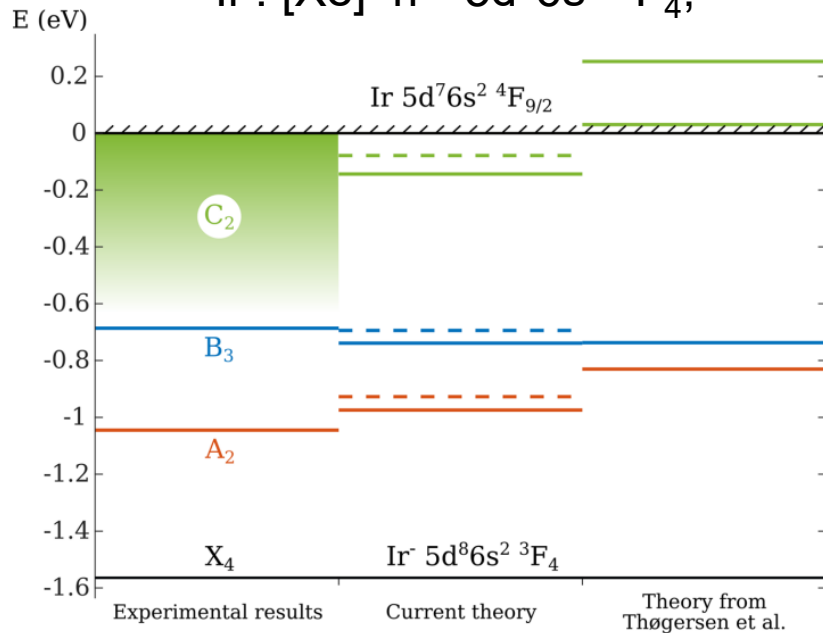
Five elements with
allowed transitions
found

Some lifetimes of
metastable states have
been determined



Essentially no traditional optical spectroscopy!

Ir: [Xe] 4f¹⁴ 5d⁸6s² ³F₄,



State	τ_{MCDHF}	τ_{expt}
A_2	> 2500 s	> 1200 s
B_3	112(11) ms	133(10) ms
C_2	161(8) ms	172(35) ms

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Experimental and theoretical studies of excited states in Ir⁻

M. K. Kristiansson^{1,*}, S. Schiffmann^{2,3}, J. Grumer⁴, J. Karls⁵, N. de Ruette^{1,6}, G. Eklund¹, V. Ideböhn⁵, N. D. Gibson⁷, T. Brage², H. Zettergren¹, D. Hanstorp⁵ and H. T. Schmidt¹

¹Department of Physics, Stockholm University, AlbaNova, SE-106 91 Stockholm, Sweden

²Department of Physics, Division of Mathematical Physics, Lund University, Box 118, SE-221 00 Lund, Sweden

³Spectroscopy, Quantum Chemistry and Atmospheric Remote Sensing (SQUARES), CP160/09, Université Libre de Bruxelles (ULB), B 1050 Bruxelles, Belgium

⁴Department of Physics and Astronomy, Theoretical Astrophysics, Uppsala University, Box 516, SE-751 20 Uppsala, Sweden

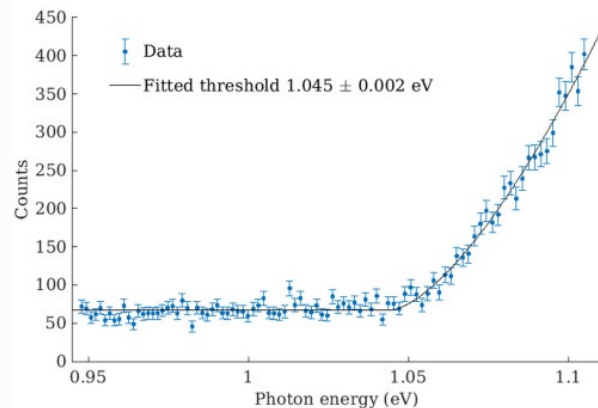
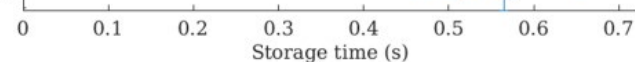
⁵Department of Physics, University of Gothenburg, SE-412 96 Gothenburg, Sweden

⁶European Spallation Source ERIC (ESS), SE-221 00 Lund, Sweden

⁷Department of Physics and Astronomy, Denison University, Granville, Ohio 43023, USA



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Experimental and theoretical studies of excited states in Ir^-

M. K. Kristiansson^{1,*}, S. Schiffmann^{2,3}, J. Grumer⁴, J. Karls⁵, N. de Ruette^{1,6}, G. Eklund¹, V. Ideböhn⁵,
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¹*Department of Physics, Stockholm University, AlbaNova, SE-106 91 Stockholm, Sweden*

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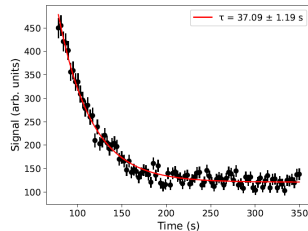
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⁷*Department of Physics and Astronomy, Denison University, Granville, Ohio 43023, USA*

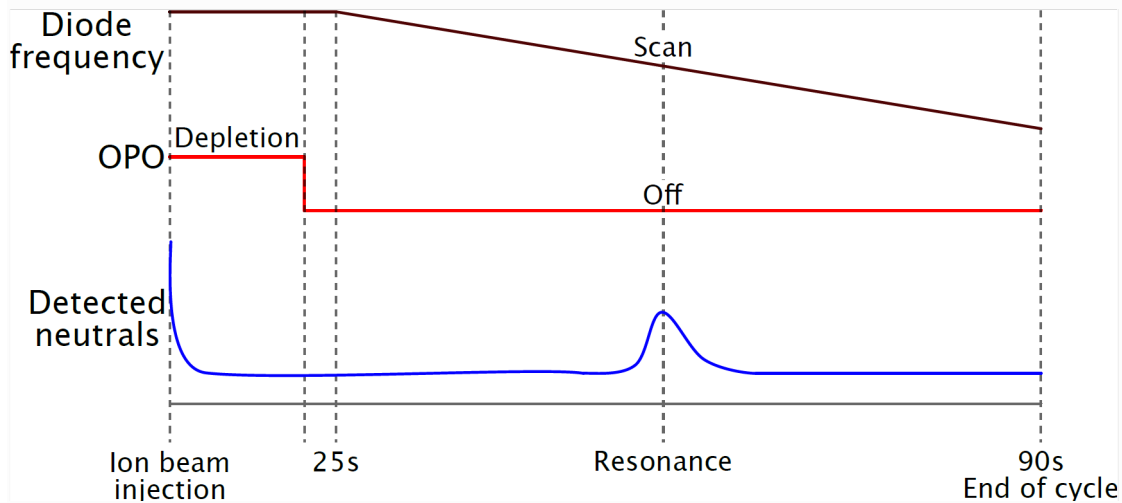
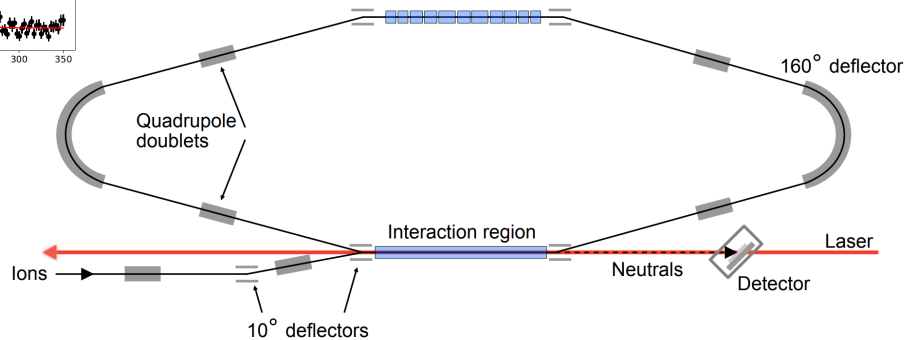
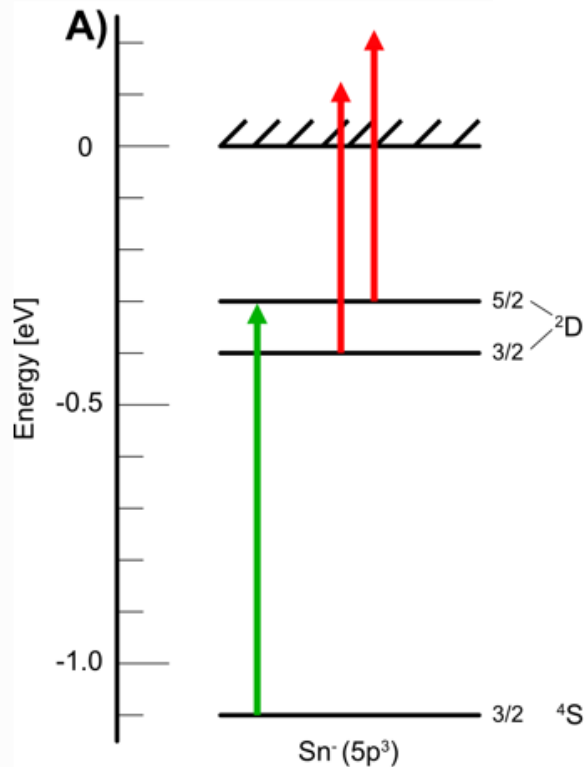


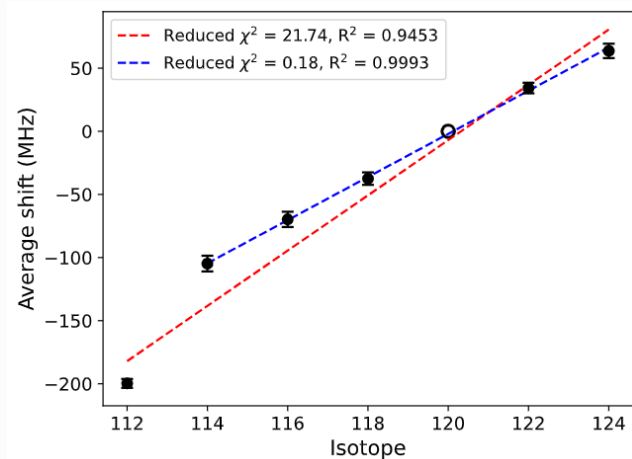
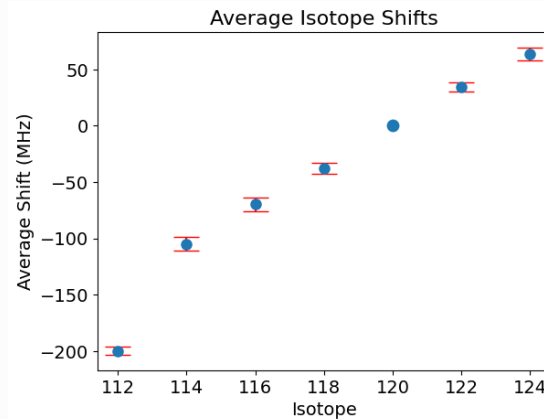
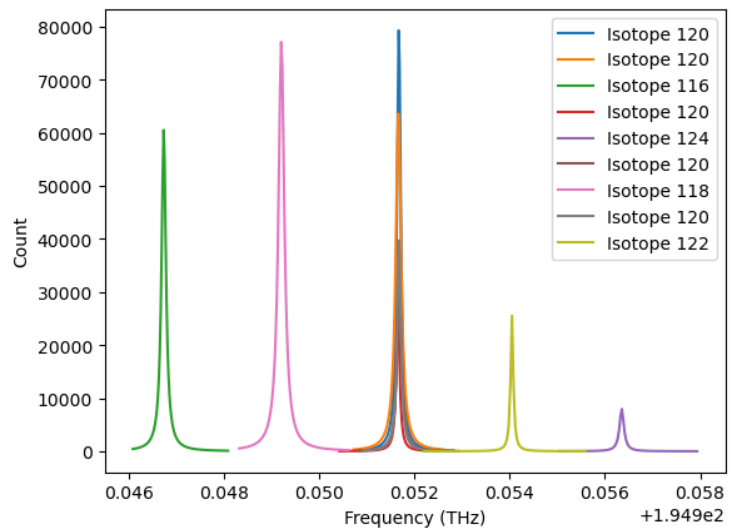
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Sn⁻



3





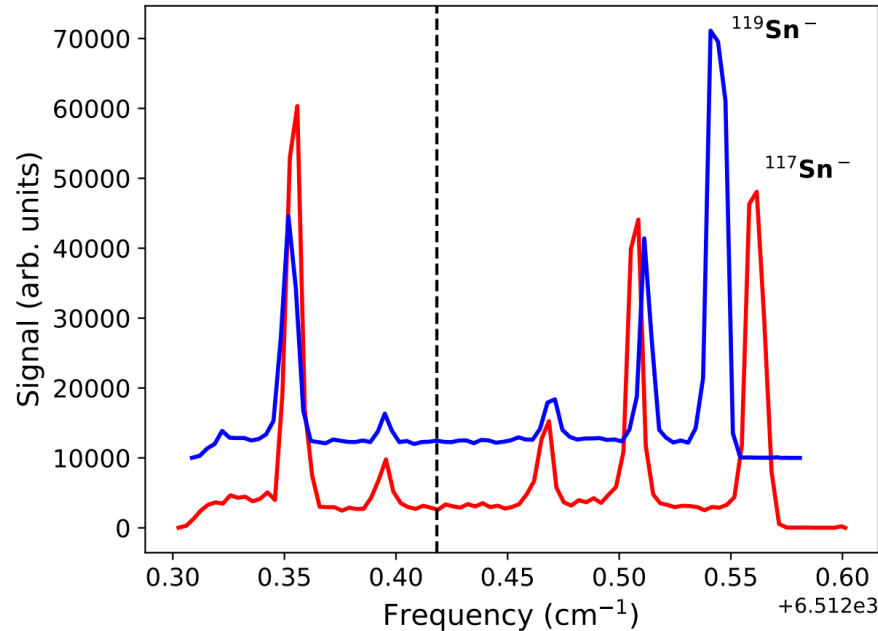
$$\Delta\nu^{A,A'} = \nu^{A'} - \nu^A = k_{\text{MS}} \frac{m_{A'} - m_A}{m_{A'} m_A} + F \delta \langle r^2 \rangle^{A,A'}$$

Transition	F	$F_{\text{exp.}}$	k_{NMS}	k_{SMS}	$k_{\text{SMS,exp.}}$	k_{MS}	$k_{\text{MS,exp.}}$
$^4S_{3/2} \rightarrow ^2D_{3/2}$	0.26	-	98.3	-181.3	-	-83.0	-
$^4S_{3/2} \rightarrow ^2D_{5/2}$	0.24	-0.03	107.5	-44.5	164.8	63.0	272.3

Theory made by A. Bondarev

[45] M. S. Safronova, M. G. Kozlov, W. R. Johnson, and D. Jiang, [Phys. Rev. A 80, 012516 \(2009\)](#).

Hyperfine structure



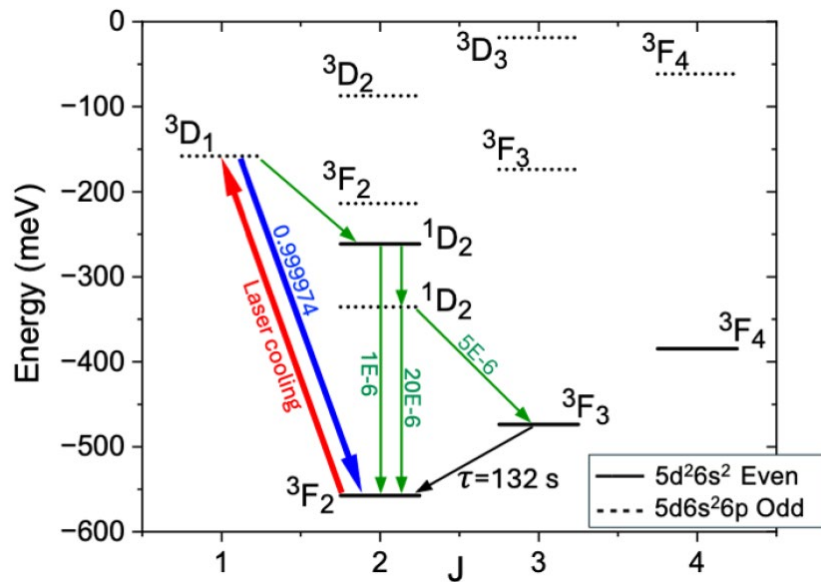
$I = 1/2$ for both ^{117}Sn and ^{119}Sn

→ Should have 4 HFS components

*Magnetic moments for
 ^{117}Sn and ^{119}Sn are very similar*

Forbidden transitions are not forbidden anymore!

La⁻



CI+all-order

- [24] M. S. Safronova, M. G. Kozlov, W. R. Johnson, and D. Jiang, Phys. Rev. A **80**, 012516 (2009).
- [25] C. Cheung, M. G. Kozlov, S. G. Porsev, M. S. Safronova, I. I. Tupitsyn, and A. I. Bondarev, Computer Physics Communications **308**, 109463 (2025).

TABLE I. Measured and calculated lifetimes of La⁻ excited states in seconds.

State	Present Experiment	Present Theory	O'Malley [14]	Cerchiari [11]
$^3F_3^e$	132(20)	131	132*	132
$^3F_4^e$	32(5)	135	134*	
Short-lived	0.22(3)			

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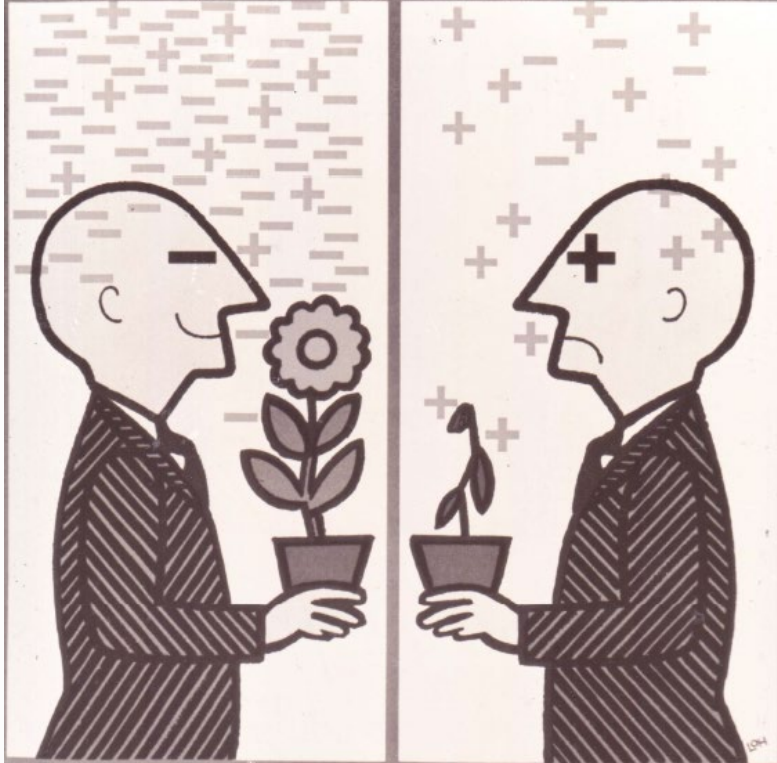
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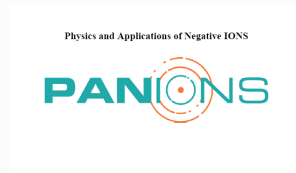
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FÖR VETENSKAPLIG FORSKNING